

Deep ocean ventilation, carbon isotopes, marine sedimentation

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# Deep ocean ventilation, carbon isotopes, marine sedimentation and the deglacial CO<sub>2</sub> rise

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## Abstract

The link between the atmospheric CO<sub>2</sub> level and the ventilation state of the deep ocean is an important building block of the key hypotheses put forth to explain glacial-interglacial CO<sub>2</sub> fluctuations. In this study, we systematically examine the sensitivity of atmospheric CO<sub>2</sub> and its carbon isotope composition to changes in deep ocean ventilation, the ocean carbon pumps, and sediment formation in a global three-dimensional ocean-sediment carbon cycle model. Our results provide support for the hypothesis that a break up of Southern Ocean stratification and invigorated deep ocean ventilation were the dominant drivers for the early deglacial CO<sub>2</sub> rise of ~35 ppm between the Last Glacial Maximum and 14.6 ka BP. Another rise of 10 ppm until the end of the Holocene is attributed to carbonate compensation responding to the early deglacial change in ocean circulation. Our reasoning is based on a multi-proxy analysis which indicates that an acceleration of deep ocean ventilation during the early deglaciation is not only consistent with recorded atmospheric CO<sub>2</sub> but also with the reconstructed opal sedimentation peak in the Southern Ocean at around 16 ka BP, the record of atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$ , and the reconstructed changes in the Pacific CaCO<sub>3</sub> saturation horizon.

## 1 Introduction

Ancient air samples trapped in polar ice reveal that atmospheric CO<sub>2</sub> has undergone large natural fluctuations with an amplitude of 90–100 ppm between glacial and interglacial periods. Antarctic temperature closely parallels the recorded atmospheric CO<sub>2</sub> concentration during at least the last eight glacial cycles (e.g. Lüthi et al., 2008).

The global oceans have been identified as the key regulator of the atmospheric carbon content on glacial-interglacial time-scales (e.g. Sigman and Boyle, 2000; Archer et al., 2000; Siegenthaler and Wenk, 1984; Sarmiento and Toggweiler, 1984; Knox and McElroy, 1984). The ocean carbon cycle is driven by a multitude of different processes such as ocean circulation, the ocean carbon pumps and marine sediment formation

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which equilibrate with atmospheric CO<sub>2</sub> on time-scales of months to millennia. The main focus of this study is to examine how sensitively atmospheric CO<sub>2</sub> is linked to the ventilation rate of the deep ocean watermasses under changes in ocean-internal carbon cycling and sediment formation.

5 We perform a series of illustrative sensitivity experiments with the coupled Bern3D+C ocean-sediment carbon cycle model in which the rate of deep ocean ventilation is systematically varied by changing wind forcing. The time-dependent response for multiple carbon cycle proxies such as  $\delta^{13}\text{C}$ - and  $\Delta^{14}\text{C}$ -signatures in the ocean and the atmosphere but also the depth of the CaCO<sub>3</sub> saturation horizon and the sedimentation rate of CaCO<sub>3</sub>, organic carbon and opal are compared to available paleo-data  
10 to evaluate the importance of changes in deep ocean ventilation for glacial-interglacial CO<sub>2</sub> variations.

Paleo-records of the CO<sub>2</sub> carbon isotope composition provide valuable insight into past sources and sinks of the atmospheric carbon reservoir. This information can be  
15 used to constrain contributions from the individual processes made responsible for the CO<sub>2</sub> fluctuations (e.g. Elsig et al., 2009; Joos et al., 2004; Hughen et al., 2004; Smith et al., 1999; Leuenberger et al., 1992). Unfortunately, during glacial-interglacial transitions there were generally several simultaneous processes generating similar imprints on CO<sub>2</sub>,  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$ . So far, the isotope records of atmospheric CO<sub>2</sub> alone  
20 have not allowed for a unique quantitative process identification (Köhler et al., 2006).

The timing and coevolution of the CO<sub>2</sub> carbon isotope records over the last deglaciation nevertheless indicate that the initial CO<sub>2</sub> increase at the beginning of deglaciations was caused by the injection of some old carbon stock with low  $\delta^{13}\text{C}$  signature into the atmosphere as both records of atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  exhibit a relatively  
25 rapid decline during the early deglaciation. Such a carbon release could in principle come from terrestrial or oceanic sources. Zeng (2003) proposes that there is a 547 GtC carbon release from glacial to interglacial originating from organic carbon under the great ice sheets of the Northern Hemisphere. However, the observed atmospheric CO<sub>2</sub> variations are not in phase with reconstructed sea level fluctuations. Recorded

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atmospheric CO<sub>2</sub> concentrations rather show an extraordinary correlation with Southern Ocean (SO) climate changes. It thus appears more plausible that the initial driver of the deglacial CO<sub>2</sub> rise is located in the Southern Ocean (e.g. Fischer et al., 2009).

There is evidence that low glacial CO<sub>2</sub> concentrations can at least partly be attributed to an increased isolation of deep water masses from the atmosphere (e.g. Skinner et al., 2010; Boyle, 1988). Many authors postulate that during glacial times there existed a relatively large and poorly ventilated abyssal water mass which led to a relocation of carbon from the atmosphere to the deep ocean. As most deep ocean watermasses are ventilated through the SO surface (Primeau, 2005), physical processes in this region are put forth as the responsible drivers of the postulated changes in deep ocean overturning (e.g. Fischer et al., 2009; Sigman and Boyle, 2000). It has been proposed that variations in deep ocean ventilation have resulted from (i) changes in SO wind forcing (Toggweiler et al., 2006), (ii) changes in SO buoyancy forcing (Watson and Naveira-Garabato, 2006) or (iii) changes in SO water column stratification (Schmittner and Galbraith, 2008; Francois et al., 1997). Nevertheless, no consensus has emerged so far neither on the physical driving forces of the ventilation changes nor on the magnitude of the associated CO<sub>2</sub> variations.

Reconstructions of the glacial  $\delta^{13}\text{C}$  distribution from benthic foraminifera in the Pacific Ocean show that – relative to the Holocene – deep water below 2000–2500 m was more depleted in  $\delta^{13}\text{C}$  compared to the upper water mass (Matsumoto et al., 2002; Keigwin, 1998). In the Indian, Atlantic and Southern Oceans, reconstructed glacial-age vertical  $\delta^{13}\text{C}$  gradients point at the existence of a sharp chemocline at similar depths separating the well-ventilated water above from poorly ventilated water below (e.g. Hodell et al., 2003; McCorkle et al., 1998; Curry et al., 1988; Duplessy et al., 1988). These  $\delta^{13}\text{C}$ -data support the “nutrient deepening” hypothesis (Boyle, 1988) calling for a transfer of nutrients and DIC from upper- and mid-depth waters to the deep ocean during glacial times. According to this scenario, the main part of the glacial CO<sub>2</sub> drawdown is the result of a more efficient marine biological pump in response to a poorly ventilated deep ocean.

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Further evidence for a slowed deep ocean ventilation during glacial periods comes from recorded variations of opal burial in SO sediments (Anderson et al., 2009) and the occurrence of  $\delta^{13}\text{C}$  carbon isotope minima in planktic foraminifera at the beginning of glacial terminations (Spero and Lea, 2002). These signals have been interpreted to reflect an invigoration in SO ventilation during the last glacial-interglacial transition. As a consequence, the breakdown of surface water stratification and renewed Circumpolar Deep Water (CDW) upwelling in the SO would bring the postulated abyssal watermass – which was isolated from the atmosphere during the glacial period – to the ocean surface.

The large recorded drop in atmospheric  $\Delta^{14}\text{C}$  of about 300‰ over the course of the last deglaciation may further support this interpretation as the drop cannot be attributed solely to a decrease in stratospheric  $^{14}\text{C}$ -production (Hughen et al., 2004; Muscheler et al., 2004). Radiocarbon measurements on glacial-age sediments from the North Pacific (Galbraith et al., 2007; Marchitto et al., 2007) and from the deep Atlantic (Keigwin and Schlegel, 2002; Boyle, 1988) indicate that abyssal waters were indeed less ventilated during glacial times. However, surface-to-deep  $^{14}\text{C}$ -age differences reconstructed from a sediment core in the western equatorial Pacific show that late-glacial ventilation at a depth of 2.8 km was not significantly lower than today (Broecker et al., 2007). Summarizing the available evidence from carbon isotopes, Broecker and Barker, 2007 come to the conclusion that the volume of an isolated abyssal reservoir during glacials is limited to no more than half of the ocean.

The “shelf hypothesis” (Broecker, 1982) is an alternative oceanic mechanism that has been put forth to explain the recorded glacial-interglacial  $\text{CO}_2$  fluctuations. This mechanism relies on changes in the biological pump strength driven by shifts in organic matter accumulation on shallow shelves as sea level rises and falls. Atmospheric  $p\text{CO}_2$  would be reduced during glacial times through increased biological productivity promoted by an enhanced whole-ocean nutrient inventory. Paleo-data have revealed, however, that several prerequisites are not met which are required for the shelf hypothesis to explain the full glacial  $\text{CO}_2$  drawdown: For instance, there is no sufficiently

large amount of organic sediments on shelf areas to provide additional nutrients during glacials (Peacock et al., 2006).

Besides reorganizations of the ocean-internal carbon cycle and the terrestrial biosphere, transient imbalances between marine sediment formation and continental weathering fluxes may also have contributed significantly to the recorded millennial-scale variations in atmospheric CO<sub>2</sub>, δ<sup>13</sup>C<sub>CO<sub>2</sub></sub> and Δ<sup>14</sup>C<sub>CO<sub>2</sub></sub> (e.g. Archer et al., 2000). Most prominently, the mechanism of carbonate compensation may have accounted for up to a third of the glacial-interglacial CO<sub>2</sub> change without conflicting with reconstructed seafloor CaCO<sub>3</sub> preservation patterns (Marchitto et al., 2004). Imbalances in the marine organic matter budget have received less attention in previous studies. Nevertheless, there is a priori no reason why excess burial or excess weathering of organic matter should not also have affected atmospheric CO<sub>2</sub> levels in the past.

Numerical models allow a process-based interpretation of the recorded CO<sub>2</sub> variations and to quantitatively test different scenarios of glacial-interglacial CO<sub>2</sub> variations. A dilemma is that the application of comprehensive climate-carbon models for multi-millennial simulations and for exploring the sensitivity of results to uncertain parameters is restricted by their large CPU-demand. On the other hand, box models are inherently limited as they do not include ocean dynamics and only poorly represent the three-dimensional geometry of the ocean and ocean-sediment interactions.

The goal of this study is to explore the carbon cycle implications of altered deep ocean ventilation and biological carbon pumps together with the millennial-scale effects of implied changes in marine sediment formation. For this purpose, we use the global coarse-resolution three-dimensional Bern3D+C ocean model (Müller et al., 2006; Tschumi et al., 2008) which is coupled to a spatially-resolved dynamic model for marine sediment formation (Tschumi, 2009). This modelling framework allows for the simulation of multiple tracers over long time periods of several 10 000 years which are required for sediment formation processes to relax to equilibrium.

The model simulates the cycling of total carbon (<sup>12</sup>C + <sup>13</sup>C + <sup>14</sup>C) and the carbon isotopes <sup>13</sup>C and <sup>14</sup>C between the atmosphere, the terrestrial biosphere, the ocean

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reservoir, and the marine sediments. The ocean circulation field, the ocean carbon pumps and the formation of organic, opal, and  $\text{CaCO}_3$  sediments at the seafloor are calculated from prognostic dynamical equations discretized on three-dimensional grids. The terrestrial reservoir and the atmosphere are represented rather simplistically as single well-mixed boxes. The carbon stock on land is fixed (2220 GtC), but not its isotopic composition.

As a starting point of this study, we calculate the response of the coupled model to a change in the export ratio of organic matter to calcite (Sigman et al., 1998). This experiment will provide a test for the model representation of carbonate compensation and allow to compare our modelling framework to others. The second model experiment is an increase of the whole-ocean nutrient inventory by 30% (Sigman et al., 1998; Broecker and Peng, 1987) serving as a test for the “shelf hypothesis”. In contrast to previous similar experiments, in our model enhanced export production of organic matter will eventually relax back to initial rates as excess organic matter sedimentation removes from the ocean the injected nutrients in the long term. Therefore, the modeled shift in the Pacific  $\text{CaCO}_3$  saturation horizon in our experiment turns out to be small as opposed to previous studies (Heinze et al., 1991; Sigman et al., 1998). Finally, but as the main interest of this study, we probe the response of variations in ocean deepwater ages driven by changes in SO wind forcing and the implied changes in SO deep convection for a range of different ocean ventilation time-scales.

The novel aspect of our analysis is the simulation of the time-dependent response to prototype scenarios considering  $^{13}\text{C}$ - and  $^{14}\text{C}$ -cycling explicitly in a fully-coupled dynamical ocean-sediment model with sedimentation of all particulate species, i.e. the marine organic matter,  $\text{CaCO}_3$  and opal cycles are all represented as “open cycles” in this study. The broad range of processes and variables simulated increases the complexity of the ocean carbon cycle in the model, but it also allows for a more robust simultaneous comparison of the model results with multiple proxies. The systematic variations in deep ocean water ages through changes in the wind forcing and deep convection in the Southern Ocean represents a new approach to explore the sensitivity

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of atmospheric CO<sub>2</sub> to deep ocean ventilation and marine sediment formation in three-dimensional models.

## 1.1 Model description

The version of the Bern3D+C carbon cycle model used here comprises an atmospheric, a terrestrial, an oceanic and a sedimentary reservoir which are coupled through the exchange of carbon and its isotopes <sup>13</sup>C and <sup>14</sup>C. The model's ocean physics is described in detail in Müller et al. (2006), marine biogeochemical cycles in Parekh et al. (2008) and Tschumi et al. (2008) and the coupling to the sedimentary component in Tschumi (2009). Unless explicitly stated otherwise, the model is run under standard parameters as mentioned in these references.

Marine biogeochemical cycles and air-sea gas-exchange are implemented following OCMIP-2 (Orr and Najjar, 1999; Najjar et al., 1999) with the addition of prognostic formulations for marine biological productivity as well as representations for the cycling of iron (Parekh et al., 2008), silica (Tschumi et al., 2008), <sup>13</sup>C and <sup>14</sup>C.

Export production is a function of temperature, light, phosphate and iron following Doney et al. (1996). The competition between opal and calcite producers is modeled following formulations of the HAMOCC5 model of Maier-Reimer et al. (2005). The half-saturation constant for silica-uptake is a function of local silica concentration and the ratio of silica to phosphorous uptake by opal producers varies with silica, light, iron and phosphate following Aumont and Bopp (2006).

The atmosphere is considered as a single well-mixed box. To calculate air-sea gas-exchange, a constant value for the square of the wind-speed  $\overline{u^2}$  is employed which is set equal to the spatio-temporal mean of the climatology for  $u^2$  provided by OCMIP-2 (Najjar et al., 1999).

The land biosphere model is taken from Siegenthaler and Oeschger (1987) which is composed of 4 well-mixed compartments with fixed carbon exchange fluxes and stocks: ground vegetation and leaves (100 GtC), wood (500 GtC), detritus (120 GtC) and soils (1500 GtC).

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The sedimentary component is based on the model described by Gehlen et al. (2006) and Heinze et al. (1999) which represents sediment diagenesis in the top 10 cm beneath seafloor. Sediment formation, i.e. the accumulation of opal,  $\text{CaCO}_3$ , organic matter and clay sediments is calculated on the basis of a set of dynamical equations for the sedimentary diagenetic processes. The redissolution and remineralization reactions all follow first order kinetics. Two types of remineralization reactions are considered, oxic respiration and denitrification. For more details on the sediment model component and model equations refer to the appendix.

In the model, cycling of  $^{13}\text{C}$  is subject to the following fractionation effects: In the ocean, equilibrium fractionation is considered in the chemical carbonate equilibrium system (Mook, 1988). For air-sea gas-exchange, a kinetic fractionation is computed (Siegenthaler and Muennich, 1981). Further fractionations occur during marine photosynthesis (Freeman and Hayes, 1992) and the formation of calcium carbonate (Mook, 1988). In the land biosphere, isotopic fractionation is accounted for during photosynthesis.

Cycling of  $^{14}\text{C}$  is simulated explicitly, i.e. the actual concentrations of  $^{14}\text{C}$  are transported and not the  $^{14}\text{C}/^{12}\text{C}$ -ratio. In contrast to  $^{13}\text{C}$ , cycling of  $^{14}\text{C}$  is not affected by fractionation effects in the model. This allows to directly infer  $\Delta^{14}\text{C}$  values from the modeled distribution without performing a fractionation correction.

### 1.2 Model spin-up and preindustrial steady state

The coupled carbon cycle model was spun up by bringing the terrestrial biosphere, the ocean and the sediments into steady state under fixed atmospheric values for  $p\text{CO}_2$  (278.00 ppm),  $\delta^{13}\text{C}$  ( $-6.305\text{‰}$ ) and  $\Delta^{14}\text{C}$  (0.00‰). A fixed production rate for  $^{14}\text{C}$  is then prescribed and  $p\text{CO}_2$ ,  $\delta^{13}\text{C}$  and  $\Delta^{14}\text{C}$  are let evolve dynamically in the atmosphere to perform the subsequent sensitivity simulations. The remaining drifts in the control simulation amount to less than  $3 \times 10^{-3}$  ppm/kyr in  $p\text{CO}_2$ , less than  $2 \times 10^{-5}\text{‰/kyr}$  in  $\delta^{13}\text{C}$  and less than  $3 \times 10^{-3}\text{‰/kyr}$  in  $\Delta^{14}\text{C}$ . As the typical model response to the perturbations applied is much larger than that, the spun up model state is sufficiently close to equilibrium for our purposes.

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The combined ocean/sediment compartment is separately initialized as follows before it is coupled interactively to the atmosphere. To spin up ocean physics in a first step, sea surface temperatures (SST) and salinities (SSS) are restored towards modern monthly observations (Levitus et al., 1994; Levitus and Boyer, 1994) and wind stress is prescribed according to the monthly climatology from NCEP/NCAR (Kalnay and et al., 1996). After 10 000 model years, mixed boundary conditions are imposed for SST and SSS following Tschumi et al. (2008). Inspection of the modeled ocean circulation field and the temperature and salinity distributions at steady state reveal that the primary flow paths of the global ocean circulation and the important water masses are represented adequately (see Müller et al., 2006 and Tschumi et al., 2008). The annual mean maximum Atlantic meridional overturning circulation (AMOC) amounts to 14.7 Sv, comparing favorably to the estimate of  $17 \pm 5$  Sv (Talley et al., 2003).

In a second step, the steady state ocean circulation is then used to spin up marine biogeochemical cycles during 10 000 model years. The resulting steady state fields of the modeled biogeochemical tracers successfully reproduce the large-scale patterns of the observed distributions in the modern ocean (see Tschumi et al., 2008; Müller et al., 2006 and Fig. 3). Simulated export fluxes for particulate organic matter ( $12.76 \text{ GtC yr}^{-1}$ ),  $\text{CaCO}_3$  ( $1.13 \text{ GtC yr}^{-1}$ ) and opal ( $95 \text{ Tmol Si yr}^{-1}$ ) all lie within available estimates (Jin et al., 2006; Sarmiento and Gruber, 2006).

In a next phase, the sediments are “filled up” during 50 000 years by imposing spatially uniform, but temporally variable weathering fluxes to the sea surface for phosphate, silicic acid, alkalinity, carbon,  $^{13}\text{C}$  and  $^{14}\text{C}$ . The weathering rates are set equal to the net flux of the respective tracers across the ocean-sediment interface. As the sediment inventories converge to their steady state values, weathering rates stabilize to balance with loss of solid matter that results from sediment burial. These values are diagnosed and prescribed subsequently as temporally constant weathering input. The ocean-sediment component is run into a new equilibrium during 30 000 years in this new setup.

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Simulated sediment accumulation amounts to 0.14 GtC/yr for organic matter and to 0.17 GtC/yr for CaCO<sub>3</sub>. These rates are roughly compatible with data-based estimates of 0.12–0.26 GtC/yr for organic matter burial and of 0.1–0.14 GtC/yr for CaCO<sub>3</sub> burial (Sarmiento and Gruber, 2006; Feely et al., 2004). In a series of carbon pulse release experiments, CO<sub>2</sub> neutralization time scales were found to be between 5–8 kyr for seafloor CaCO<sub>3</sub> neutralization and between 8–13 kyr for terrestrial CaCO<sub>3</sub> neutralization (Tschumi, 2009). These values are compatible with previous studies (e.g. Archer et al., 1997). The model equations, the model parameters and the details of the ocean-sediment coupling can be found in Tschumi (2009).

Simulated sediment composition in the coupled steady state is shown in Fig. 1a, b and c.

## 2 Model experiments and results

### 2.1 Rain ratio reduction

The marine biological pumps (Volk and Hoffert, 1985), i.e. the organic matter and the carbonate pumps, are central drivers of the ocean carbon cycle. For two separate reasons, they are of great importance for the regulation of the atmospheric CO<sub>2</sub> level by the oceans. First, they establish systematic gradients in dissolved inorganic carbon (DIC) and alkalinity between the surface and the deep ocean. The overall impact of both pumps together are lower *p*CO<sub>2</sub> levels in the surface ocean compared to an abiotic ocean. Second, accumulation of organic matter and CaCO<sub>3</sub> sediments is driven by the deposition of biogenic material on the seafloor. Any change in the marine biological pumps thus also affects organic matter and CaCO<sub>3</sub> sedimentation with potentially significant impacts on atmospheric CO<sub>2</sub> through perturbations of the marine organic matter and CaCO<sub>3</sub> balances.

The ocean carbonate pump is driven by the production of biogenic CaCO<sub>3</sub> shells in the sea surface. The sinking CaCO<sub>3</sub> shells entrain alkalinity to the deep ocean

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and hence increase sea surface  $p\text{CO}_2$  by shifting the speciation of surface ocean DIC towards dissolved  $\text{CO}_2$ . A reduced export ratio between  $\text{CaCO}_3$  and organic matter export (rain ratio) would reduce sea surface  $p\text{CO}_2$  not only due to an ocean-internal redistribution of alkalinity but also as a result of a lower carbonate concentration in the deep ocean triggering redissolution of marine  $\text{CaCO}_3$  sediments. The resulting imbalance between the weathering flux of dissolved  $\text{CaCO}_3$  into the ocean and marine  $\text{CaCO}_3$  sedimentation would drive a rise in the ocean's total alkalinity and additionally contribute to lower sea surface  $p\text{CO}_2$ . Carbonate compensation is the feedback mechanism responsible for re-establishing the ocean  $\text{CaCO}_3$  balance on a time-scale of order 10 kyr through shifts in the depth of the  $\text{CaCO}_3$  saturation horizon (Archer et al., 2000).

By reason of their central role in the regulation of the ocean's alkalinity budget rain ratio mechanisms were among the first candidates made responsible to drive glacial-interglacial  $\text{CO}_2$  fluctuations in the atmosphere (e.g. Archer et al., 2000; Berger and Keir, 1984). A reduced rain ratio during glacial times may have resulted for instance from increased diatom productivity in lower latitudes at the expense of coccolithophorids (silica leakage hypothesis, Matsumoto et al., 2002) or from relaxed nutrient-limitation in the sea surface (Margalef, 1978). The inferred  $\text{CaCO}_3$  preservation history recorded in marine sediments, however, constrains the contribution of whole-ocean alkalinity changes to roughly a third of glacial-interglacial  $\text{CO}_2$  variations (Marchitto et al., 2004).

In our first model experiment, we attempt to disentangle and quantify the complex response in ocean biogeochemistry to a reduced rain ratio. We perform parallel simulations in a coupled ocean-sediment setup and in an ocean-only setup to separate ocean-internal mechanisms from interactions with the sediments. Following Sigman et al. (1998), we abruptly reduce  $\text{CaCO}_3$  export by 20% uniformly in the global sea surface. This corresponds in our model to a decline in the global rain ratio from 0.088 to 0.071.

## 2.1.1 Results

Figure 3 shows the time-dependent carbon cycle responses as simulated with the ocean-only model (black lines) and the ocean-sediment model (red lines). In response to decreased  $\text{CaCO}_3$  export atmospheric  $\text{CO}_2$  drops from 278 ppm to 268 ppm within centuries (e-folding time scale: 390 yrs) in the ocean-only model (Fig. 3a, black line). Mean surface alkalinity rises as less  $\text{CaCO}_3$  shells sink from the surface to the ocean interior (Fig. 3b, black line). Reduced alkalinity in the ocean interior in turn lowers  $[\text{CO}_3^{2-}]$  pushing the  $\text{CaCO}_3$  saturation horizon upwards by approximately 300 m (Fig. 3d, black line).

In the ocean-sediment model the  $\text{CO}_2$  drawdown is strongly amplified to 237 ppm due to the increase in whole-ocean alkalinity (Fig. 3a and b, red line). Carbonate compensation responds rather slowly with an e-folding time scale of 11.7 kyr. Reduced  $\text{CaCO}_3$  export and deposition on the seafloor drive a deepening in the  $\text{CaCO}_3$  saturation horizon as to provide a larger area on the seafloor where the rain of  $\text{CaCO}_3$  is efficiently preserved (Fig. 3d, red line). As a consequence, the simulated  $\text{CaCO}_3$  sedimentation rate starts to recover after the initial abrupt decline in order to restore the ocean's  $\text{CaCO}_3$  balance in the model (Fig. 3c). The final total shift in steady state saturation depth in the ocean-sediment model amounts to a deepening of 700 m in the Atlantic and of 800 m in the Pacific basin (Fig. 3d, red line).

The simulated responses in atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  to changes in the rain ratio are small. The ocean-internal distribution of  $\delta^{13}\text{C}_{\text{DIC}}$  and  $\Delta^{14}\text{C}_{\text{DIC}}$  is only marginally affected by the marine  $\text{CaCO}_3$  cycle. In the ocean-only experiment, we therefore simulate negligible changes in atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  as  $\text{CaCO}_3$  export is reduced. In fact, there is virtually no response in atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  (Fig. 3e, black line) and  $\Delta^{14}\text{C}_{\text{CO}_2}$  is raised by a modest 3‰ (Fig. 3f, black line).

In the coupled ocean-sediment experiment, the system response in carbon isotope signatures of atmospheric  $\text{CO}_2$  is additionally sensitive to the decline in marine  $\text{CaCO}_3$

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sedimentation. However, the responses of  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  to carbonate compensation are rather slow and remain small (Fig. 3e and f, red lines).

### 2.1.2 Comparison to other studies

Heinze et al. (1991) have simulated a relatively modest sensitivity of atmospheric  $\text{CO}_2$  to changes in the rain ratio: They have found a  $\text{CO}_2$  drawdown of 28.5 ppm when the rain ratio is reduced by 50% in the three-dimensional Hamburg Carbon Cycle Model which was coupled to a simplistic model for sediment diagenesis with fixed sediment re-suspension rates. Archer and Maier-Reimer (1994) have shown that including organic-carbon-driven dissolution of  $\text{CaCO}_3$  sediments substantially increases the  $p\text{CO}_2$  response to rain ratio changes in their model. They reach glacial  $\text{CO}_2$  levels by reducing the rain ratio by 40% which amounts to a comparable  $\text{CO}_2$  sensitivity as in this study.

Sigman and Boyle (2000) have reduced the ratio of organic matter to  $\text{CaCO}_3$  export in the low latitudes by 50% in the CYCLOPS box model. They obtain a drawdown of 22 ppm in the ocean-only model and of 63 ppm in the ocean-sediment model.  $\text{CaCO}_3$  compensation amplifies the  $\text{CO}_2$  reduction by a factor of three in their study compared to a factor of four in this study. The reason for the comparatively lower impact of  $\text{CaCO}_3$  compensation in the box model study is probably the shorter relaxation time of  $\text{CaCO}_3$  burial. The original rate is restored after about 15 000 years, whereas in the Bern3D+C model it takes roughly 50 000 years. The longer relaxation time in our study results in a larger imbalance between  $\text{CaCO}_3$  burial and weathering when integrated over time such that the increase in whole-ocean alkalinity is larger in response to carbonate compensation.

Chikamoto et al. (2009) have lowered the rain ratio by 25% in the High-Latitude Exchange/Interior Diffusion Advection (HILDA) model coupled to a similar dynamical sediment model as used in this study. They find a drawdown of 59 ppm in atmospheric  $\text{CO}_2$  together with a deepening of the calcite saturation horizon by 1870 m. The response time scale of atmospheric  $\text{CO}_2$  and the magnitude of the drawdown is comparable to

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the results of our study, however we find a deepening of the order of only 800 m in the  $\text{CaCO}_3$  saturation horizon.

In conclusion, it appears to be challenging to robustly quantify the effect of rain ratio changes on atmospheric  $\text{CO}_2$  by means of carbon cycle models. The simulated  $\text{CO}_2$  response depends on the specific biogeochemical model state such as the initial rain ratio or the  $\text{CaCO}_3$  throughput in the oceans as well as on the model parametrization of the ocean-sediment interactions. Our results are consistent with current understanding suggesting that carbonate compensation would amplify the  $\text{CO}_2$  response of changes in the ocean carbonate pump by a factor of as much as 3–4.

## 2.2 Strengthening of the marine biological cycle

As an alternative to a reduced rain ratio, low atmospheric  $\text{CO}_2$  during glacial periods might have resulted from more vigorous biological carbon cycling in the ocean. A strengthened marine biological cycle might be driven for instance by increased nutrient availability (Broecker and Peng, 1982), iron fertilization (Martin and Fitzwater, 1988) or changes in nitrogen cycling (Falkowski, 1997).

In the next experiment, we examine the response of atmospheric  $\text{CO}_2$ ,  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  to a stimulated ocean biology by increasing the marine nutrient inventory in the Bern3D+C model. The experiment is again performed with and without coupling to sediments to isolate the effect of ocean-sediment interactions.

Following the procedure of Archer et al. (2000); Sigman et al. (1998) and Broecker and Peng (1987), we have abruptly increased  $\text{PO}_4$  concentrations uniformly by 30% to stimulate simultaneously both the organic matter and the carbonate pumps. With our model biology the global rain ratio remained virtually unchanged. Therefore, the sedimentary response on atmospheric  $\text{CO}_2$  is dominated in our model by excess organic matter sedimentation rather than by carbonate compensation.

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## 2.2.1 Atmospheric CO<sub>2</sub>

In the ocean-only model, the augmented PO<sub>4</sub> inventory drives a persistent increase in organic matter and CaCO<sub>3</sub> export production both by roughly 11% (Fig. 4c and d, black lines). The elevated rates of organic matter and CaCO<sub>3</sub> export have opposite effects on sea surface pCO<sub>2</sub>. As the strengthened organic matter pump dominates the net ocean response atmospheric CO<sub>2</sub> gets reduced to 255 ppm with an e-folding time-scale of 470 years in the ocean-only experiment (Fig. 4a, black line).

Biological productivity augments primarily in the Atlantic and in the Indian oceans, and to a lesser degree in the Northern Pacific. There is almost no change in export production in the equatorial and South Pacific as well as in the Southern Ocean because biological productivity in these regions is limited by iron or light availability. As the concurrent changes in downward fluxes of POC and CaCO<sub>3</sub> have roughly opposing effects on [CO<sub>3</sub><sup>2-</sup>] at the depth of the saturation horizon, there is only a small shift in calcite saturation depth of less than 20 m in the ocean-only model (Fig. 4b, black line).

In the coupled ocean-sediment setup, atmospheric CO<sub>2</sub> is reduced to 232 ppm within 30 kyr (Fig. 4a, red line). Thereafter, CO<sub>2</sub> slightly increases by 2 ppm until the full balance between weathering and burial of organic matter and CaCO<sub>3</sub> is re-established after 50 ky (Fig. 4e and f). The effect of ocean-sediment interactions thus enhances the CO<sub>2</sub> drawdown in the coupled system by a factor of two.

The nutrient increase abruptly boosts the export of organic matter and CaCO<sub>3</sub> by 8.5% in the ocean-sediment model (Fig. 4c and d, red lines). Subsequently, both export rates decline and reach almost original values after 50 kyr. As both changes in CaCO<sub>3</sub> and organic matter burial are taken into consideration elevated organic matter export is not sustained in the long run here, in contrast to previous studies (Sigman et al., 1998; Broecker and Peng, 1987) which have accounted solely for changes in CaCO<sub>3</sub> sedimentation.

Excess organic matter burial draws nutrients out of the ocean to bring organic sedimentation back to balance the riverine input of nutrients. The initial increase in the

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nutrient inventory is thereby neutralized in the long run. The simulated decline in  $\text{CaCO}_3$  export also follows directly from the gradual decrease in nutrient availability. In our model biology,  $\text{CaCO}_3$  export is computed from primary productivity and is not affected by changes in the calcite saturation state. The effect of  $\text{CaCO}_3$  compensation on restoring  $\text{CaCO}_3$  burial is therefore only of secondary importance, in contrast to the rain ratio reduction experiment.

The modeled  $\text{CaCO}_3$  saturation horizon deepens by about 150 m in the Atlantic and by 300 m in the Pacific during the first 15 000 years with a subsequent shoaling in both basins thereafter (Fig. 4b, red line). The first part of the response is dominated by excess POC burial, the second part by excess  $\text{CaCO}_3$  burial. Since excess POC burial implies a loss of whole-ocean DIC, it tends to push the saturation horizon downwards. On the contrary, excess  $\text{CaCO}_3$  burial leads to a shoaling of the saturation horizon as alkalinity and DIC are removed from the ocean in a 2:1-ratio. The final shift in the calcite saturation horizon after 50 kyr is a deepening of less than 100 m.

### 2.2.2 Atmospheric $\delta^{13}\text{C}_{\text{CO}_2}$ and $\Delta^{14}\text{C}_{\text{CO}_2}$

The simulated changes in atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  are predominantly controlled by the response in organic matter cycling. In the ocean-only model, atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  rises by 0.16‰ and  $\Delta^{14}\text{C}_{\text{CO}_2}$  by 7‰ in response to the strengthened marine biological cycle (Fig. 4g and h, black lines).  $\delta^{13}\text{C}_{\text{CO}_2}$  rises as surface ocean  $\delta^{13}\text{C}_{\text{DIC}}$  is shifted towards more positive values due to enhanced production of organic matter. The increase in  $\Delta^{14}\text{C}_{\text{CO}_2}$  is the direct consequence of reduced atmospheric  $\text{CO}_2$  under a constant  $^{14}\text{C}$ -production rate.

In the ocean-sediment model  $\delta^{13}\text{C}_{\text{CO}_2}$  increases by  $\sim 0.4\text{‰}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  rises by roughly 70‰ in response to a strengthened marine biological cycle (red lines in Fig. 4g and h). The time-dependent response in the carbon isotope signatures of atmospheric  $\text{CO}_2$  is more complex than in the ocean-only system. In particular, relaxation times are much longer due to the slow ocean-sediment interactions.

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CaCO<sub>3</sub> saturation depth of 500 m, whereas here we obtain only a modest deepening.

The reason for this discrepancy is mainly the fact that POC burial is not taken into account explicitly by Sigman et al. (1998). The nutrient-feedback is thus not considered. In our simulations, the loss of nutrients through excess POC burial on the one hand draws carbon out of the ocean and on the other hand limits excess CaCO<sub>3</sub> export such that CaCO<sub>3</sub> compensation has only a minor effect opposing the CO<sub>2</sub> reduction. The impact of CaCO<sub>3</sub> compensation on atmospheric CO<sub>2</sub> is seen in a small increase occurring after 30 kyr (Fig. 4a, red line).

The following conclusions emerge. Perturbations in global burial fluxes due to ocean processes are removed on a multi-millennial time scale to re-establish the balance between removal of organic matter, CaCO<sub>3</sub>, and opal by burial and corresponding weathering inputs from rivers and winds. As a consequence of this sedimentary compensation, some proxy signals that arise from changes in the internal ocean carbon cycle are partly or completely muted with time (see also next Sect. 3.3). Differences in proxies between distant time periods, for instance between the LGM and the late Holocene, may therefore partly reflect changes in the balance of the weathering-sedimentation cycle.

Kohfeld et al. (2005) compared productivity records for the late Holocene, the Last Glacial Maximum and marine isotope stage 5a to d. They concluded that export production was globally higher during the LGM than during either stage 5a-5d or the Late Holocene. They suggest a limited role for the marine biological cycle to explain glacial-interglacial CO<sub>2</sub> variations. Our results indicate that such a conclusion should be viewed with caution as differences in proxies might also be related to imbalances in the weathering-sedimentation cycle. The role of these imbalances should also be considered when analyzing whole ocean changes in  $\delta^{13}\text{C}$  to infer variations in organic matter pools (Shackleton, 1977; Bird et al., 1994), in the atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  signature (Leuenberger et al., 1992), or the evolution of nitrogen and silicate isotope signatures.

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## 2.3 Global ocean ventilation

The last set of simulations focusses on the response of atmospheric  $\text{CO}_2$ ,  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  to changes in the rate of deep ocean ventilation. In these idealized experiments we vary deep ocean ventilation in the model by scaling the prescribed amplitude of wind stress in the Southern Ocean (south of  $51^\circ\text{S}$ ). The time-dependent response of atmospheric  $\text{CO}_2$ ,  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  is determined by the complex interplay between changes in ocean circulation, marine biogeochemical cycling and sediment diagenesis subsequent to the abrupt change in wind boundary conditions. Feedbacks mediated by SST changes and the response in terrestrial biosphere are not taken into account. The simulation set consists of separate runs in which SO wind stress is gradually varied between 20% and 180% relative to the original strength by steps of 20%.

It has been found in several global ocean models (including ours) that deep mixing in the SO is driven to a large degree by momentum transfer from SH westerly winds to the ocean surface in the latitudinal band spanned by the Drake Passage (e.g. Toggweiler et al., 2006; Menviel et al., 2008; Tschumi et al., 2008). As the SO represents one of the major routes of deep ocean ventilation, the modeled characteristics of deep ocean water masses are affected on a global scale by changes in SO deep mixing. However, the relevance of these model results with regard to the physical mechanisms controlling deep mixing in the real SO is uncertain. In particular, the role of sea-ice dynamics or of small-scale eddy flows which are not represented explicitly in most global ocean models are still elusive (e.g. Spence et al., 2009; Böning et al., 2008). For these reasons, the amplitude of SO winds in our simulations is regarded as a mere tuning knob for the circulation state in the model.

### 2.3.1 Ocean circulation response

The two most prominent features in the circulation response to SO wind variations are: (1) Changes in the frequency and depth of convective events in the SO and (2) changes in the intensity and depth of the Atlantic meridional overturning circulation (AMOC).

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Stronger winds intensify convection in the SO and strengthen the AMOC (Fig. 5b and Tschumi et al., 2008). Weaker winds induce the opposite response characterized by a more stably stratified Southern Ocean and less intense AMOC.

In case of a strongly reduced wind stress (20% or 40% of the original amplitude) the AMOC breaks down completely in the model, leaving the SO as the sole region of deep water formation. These latter circulation states are named “off-states” here, whereas the model states with active AMOC are denoted as “on-states”.

As long as the AMOC remains active, i.e. in the on-states, deep ocean ventilation scales quite homogeneously across all ocean basins with the changes in SO wind amplitude (Fig. 5a). Deep ventilation ages below 2000 m vary almost linearly with the strength of SO wind stress. Globally averaged deep ventilation ages increase by 32 years per 10%-reduction in SO wind strength. The Atlantic deep water age is slightly more sensitive (43 years/10%-reduction) than in the other basins owing to the changes in AMOC strength.

In case of the off-states the link between deep ocean ventilation ages and SO wind stress deviates from the quasi-linear relationship (Fig. 5a). In particular in the deep Atlantic, the AMOC shutdown results in a large anomalous drop of roughly 600 years in ventilation age. This model response indicates the change in end-members of Atlantic deep water subsequent to the collapse of AMOC.

To sum up, changing wind stress in our model proves itself as an effective method to gradually vary globally averaged deep water age in a range between 1580 years (180% SO wind stress) and 2100 years (20% SO wind stress). Major structural shifts in the modeled global circulation pattern occur only if SO wind stress is strongly reduced to 20% or 40% where AMOC breaks down completely.

### 2.3.2 Response to sub-millennial processes

Figure 7 shows the time-dependent response for a set of informative carbon cycle parameters. The reaction of atmospheric CO<sub>2</sub> and its isotopic composition to changes in deep ocean ventilation is characterized by a major fast response unfolding during

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approximately the first 2 kyr and a minor slow response over several 10 kyr (Fig. 7a, g and h). The fast response arises from the rearrangements in ocean-internal carbon cycling, the long-term response is driven by the slow adaptation of sediment formation to the altered biogeochemical state of the ocean. Accordingly, carbon cycling in the ocean-only model has reached a new equilibrium after 2 kyr, whereas in the ocean-sediment model sedimentation processes are still under adjustment at the same time.

Figure 6 shows the simulated changes in atmospheric  $\text{CO}_2$ ,  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  for both model setups versus the variation in deep ocean ventilation age at the time slice 2 kyr after perturbing SO wind stress forcing. The results unambiguously display that better ventilated deep ocean watermasses drive an increase in atmospheric  $\text{CO}_2$ , whereas  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  decline. Ocean-only and ocean-sediment results do not substantially differ at 2 kyr since the effect from changes in sedimentation has not yet fully materialized.

Atmospheric  $\text{CO}_2$  and its isotopic signatures follow an almost linear relationship with deep ocean ventilation age for both without and with sediments as long as there is no AMOC collapse. Linearized sensitivities for  $\delta^{13}\text{C}_{\text{CO}_2}$  (0.08‰/100 years) and for  $\Delta^{14}\text{C}_{\text{CO}_2}$  (10‰/100 years) are almost identical in both model settings (Fig. 6b and c). Changes in atmospheric  $\text{CO}_2$  are slightly larger in the ocean-sediment model (−10 ppm/100 years) than in the ocean-only model (−8 ppm/100 years) at the 2 kyr time slice (Fig. 6a).

AMOC shutdowns significantly affect atmospheric  $\text{CO}_2$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$ . The simulated changes for off-states clearly deviate from the linear relationship as displayed in Fig. 6a and 6c. The AMOC shutdown induces an anomalous drawdown in atmospheric  $\text{CO}_2$  of ~10 ppm without sediments and of ~20 ppm with sediments. The anomalous rise in  $\Delta^{14}\text{C}_{\text{CO}_2}$  is of ~10‰ in the ocean-only model and of ~15‰ in the ocean-sediment model. Atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  on the other hand is rather insensitive to the shutdown of AMOC. There is no substantial deviation from linearity in the relation between  $\delta^{13}\text{C}_{\text{CO}_2}$  with the rate of deep ocean ventilation when AMOC shuts down (Fig. 6b).

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The changes in ocean ventilation have a strong impact on the marine biological carbon pumps in our model. Enhanced (reduced) ocean ventilation implies increased (diminished) upwelling of nutrients, thus promoting (weakening) biological productivity in the surface. Note that nutrient utilization behaves conversely, i.e. nutrient utilization efficiency increases as productivity decreases parallel to slowed ventilation. In case of the on-states organic matter export at 2 kyr scales roughly linearly with ventilation rising by 2% per 100-year reduction in global deep water age. The marine silica cycle on the other hand reacts more sensitively to enhanced ventilation than to a slowdown. Opal export at 2 kyr increases by 8% upon a 100 year-reduction in global deep water age and decreases by 2.3% per 100 year-increase. The full time-dependent response of POC and opal export is shown in Fig. 7c and d.

The CO<sub>2</sub> drawdown in response to slowed ocean ventilation is driven by an interplay of biological and physical processes. Both, more efficient biological pumping and a stronger solubility pump, i.e. the alteration of the deep water's physical properties induced by the shift in the proportion of end-members, contribute to the relocation of CO<sub>2</sub> from the atmosphere to the ocean. But the biological pump strongly dominates the net CO<sub>2</sub> response subsequent to wind stress changes (Tschumi et al., 2008). Phosphate concentrations decrease and  $\delta^{13}\text{C}_{\text{DIC}}$  signatures increase in the sea surface as ocean ventilation is slowed and euphotic zone nutrients are utilized more efficiently (Table 2). After propagation of this signal across the air-sea interface  $\delta^{13}\text{C}_{\text{CO}_2}$  is shifted towards less negative values in the atmosphere. Increasing  $\Delta^{14}\text{C}_{\text{CO}_2}$  in response to reduced ventilation arises from the slowed physical transport of radiocarbon to the deep ocean which ultimately reduces ocean uptake.

A specific consequence of the Southern Ocean ventilation mechanism is that vertical  $\delta^{13}\text{C}_{\text{DIC}}$ -gradients in the Southern Ocean should have steepened during glacials (Hodell et al., 2003; Ninnemann and Charles, 2002). Our results reproduce this connection between deep mixing in the Southern Ocean and the vertical gradient in  $\delta^{13}\text{C}_{\text{DIC}}$ . Reduced deep mixing evidenced by a larger  $\Delta^{14}\text{C}_{\text{DIC}}$  difference between the surface and deep Southern Ocean is accompanied by steeper vertical gradients

in  $\delta^{13}\text{C}_{\text{DIC}}$  and phosphate. According to our simulation results, upon a reduction of 100 years in the difference between surface and deep water age (mean value below 2000 m) in the Southern Ocean the  $\delta^{13}\text{C}_{\text{DIC}}$ -gradient (difference between surface and mean signature below 2000 m in the Southern Ocean) increases by 0.16‰ and the vertical phosphate-gradient grows by  $0.13 \text{ mol/m}^3$  (Table 2).

### 2.3.3 The long-term effect of ocean-sediment interactions

The long-term millennial-scale trends in atmospheric  $\text{CO}_2$ ,  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\Delta^{14}\text{C}_{\text{CO}_2}$  arise from the slow response of sediment diagenesis to the changing biogeochemical state of the ocean. The sedimentary processes tend to restore burial rates of  $\text{CaCO}_3$ , POC and opal towards initial unperturbed values in order to re-establish the balance with the fixed weathering inputs.

The simulated variations in atmospheric  $\text{CO}_2$  are amplified by 30%–60% within 50 kyr through the net effect of the sedimentary feedbacks subsequent to ventilation changes (Table 3). The overall sedimentary  $\text{CO}_2$  response is dominated by the simulated changes in  $\text{CaCO}_3$  burial. In fact, the shifts in organic matter burial counteract the simulated  $\text{CO}_2$  variations as long as AMOC is maintained (on-states). For instance, accelerated ocean ventilation leads to a net rise in atmospheric  $\text{CO}_2$ , whereas elevated organic matter burial tends to reduce  $\text{CO}_2$  by drawing carbon out of the ocean-atmosphere system. In the on-states, the sedimentary  $\text{CO}_2$  response amplification is of 30%–40% (Table 3). In case of collapsing AMOC on the other hand,  $\text{CO}_2$  changes are reinforced by the sedimentary feedbacks by as much as 60% since both carbonate compensation from reduced  $\text{CaCO}_3$  export as well as excess POC burial driven by the reduction in deep ocean oxygen availability reinforce the  $\text{CO}_2$  drawdown from slowed ventilation.

In contrast to the previous experiment in which the whole-ocean nutrient content is enhanced, the concomitant variations in deep ocean oxygen availability subsequent to ventilation changes weaken the importance of the nutrient-feedback by decoupling the

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link between shifts in POC export and in POC sediment burial. Enhanced (decreased) ventilation promotes (reduces) remineralization within the sediments through increased (decreased) oxygen concentrations in pore waters. A specific consequence is that the changes in POC export arising from altered deep ocean ventilation are maintained to a substantial degree in the long term (Fig. 7c). The POC burial flux, however, relaxes again to initial conditions to re-establish the balance with weathering input. In conclusion, the variations in oxygen availability in the ocean driven by the ventilation changes significantly decouple the link between POC export and POC burial. This decoupling complicates the reconstruction of past POC export on the basis of sedimentary records.

The sedimentary response in atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  is dominated by the temporal evolution of POC burial. For instance, in the model states with accelerated ocean ventilation excess POC burial draws organic matter with low  $\delta^{13}$  ( $\sim -20\%$ ) out of the ocean-atmosphere system. As a consequence, the sediment response is to shift  $\delta^{13}\text{C}_{\text{CO}_2}$  towards less negative values (Fig. 7g, red and yellow lines). Excess burial of POC and  $\text{CaCO}_3$  both remove carbon with low  $\Delta^{14}\text{C}$  from the ocean-atmosphere system in response to a better ventilated deep ocean. As a result, atmospheric  $\Delta^{14}\text{C}_{\text{CO}_2}$  is driven towards slightly less negative values (Fig. 7h, red and yellow lines).

Figure 7b (red and yellow lines) displays that an accelerated ocean ventilation further induces an initial deepening in the  $\text{CaCO}_3$  saturation horizon which is offset after 1–2 kyr. The first part of the response results from the removal of carbon from the ocean interior as a result of rejuvenated deep water masses and excess POC sedimentation. The subsequent shoaling is due to the slowly responding effect of carbonate compensation to excess  $\text{CaCO}_3$  burial.

The model experiments with collapsing AMOC exhibit the strongest reduction in nutrient supply to the surface and in globally integrated POC export from the surface (blue line in Fig. 7c). Global POC export remains reduced significantly in the long run. However, subsequent to the strong initial decrease POC export in the North Atlantic eventually increases after about 200 years driven by diffusion of nutrients along isopycnals to the North Atlantic ocean surface. This response in POC export production is

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qualitatively similar to the results of Chikamoto et al. (2008) where AMOC has been shut down through a freshwater discharge in the North Atlantic. In our model we additionally find a long-term decreasing trend in global POC export after 5 kyr which is due to whole-ocean nutrient loss by excess POC burial. This nutrient-feedback has not been taken into account by Chikamoto et al. (2008).

Similarly to POC export, organic matter burial strongly declines during the first 200 years (blue line in Fig. 7e). Thereafter, the burial rate sharply recovers as a result of increasing POC export in the North Atlantic and strongly reduced deep ocean oxygen supply. After around 2500 years there is excess POC burial in spite of the reduction in global POC export production. Opal export and opal sediment burial exhibit a similar response as POC export and burial (Fig. 7d and f): The AMOC shutdown induces major shifts also in the spatial pattern of opal export and sedimentation with the North Atlantic becoming a region of deep nutrient upwelling similar to the North Pacific.

The reorganization in the marine biological carbon cycle in case of AMOC shutdowns is further reflected in the time-dependent model response for the depth of the  $\text{CaCO}_3$  saturation horizon. The AMOC collapse induces an initial shoaling of the  $\text{CaCO}_3$  saturation horizon as DIC accumulates in the deep from the slowed ventilation and reduced POC burial (Fig. 7b). Consistently, the shoaling is more pronounced in the Atlantic than in the Pacific basin. After 1–2 kyr the saturation horizon starts to deepen again in both basins as deep  $[\text{CO}_3^{2-}]$  rises from the combined effects of reduced  $\text{CaCO}_3$  burial and increasing POC burial. The final steady state shift in the  $\text{CaCO}_3$  saturation horizon is a deepening of 100–200 m in the Atlantic and of 500–600 m in the Pacific.

### 3 Discussion

Our modelling approach is associated with a number of conceptual and methodological limitations worth mentioning. First, the model scenarios are idealized and not intended to directly reflect the temporal evolution of paleo-proxy signals as recorded in climatic archives. Second, changes in deep ocean ventilation are generated by varying the

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magnitude of the wind stress in the Southern Ocean by more than a factor of two. This is a convenient technical tuning mechanism to adjust the deep ocean ventilation age in the model. However, we stress the fact that these variations in SO wind strength exceed the range of realistic changes. The atmospheric CO<sub>2</sub> response to realistic variations has been found to be rather modest (Tschumi, 2009; Menviel et al., 2008).

The model simulations have been tailored to investigate the specific response of atmospheric CO<sub>2</sub> and other paleo-proxies to uniform variations in deep ocean ventilation age. Several possibly important processes to explain the full glacial-interglacial CO<sub>2</sub> variability have not been taken into account. We have neglected for instance the well-recorded variations of AMOC (McManus et al., 2004), changes in the terrestrial biosphere, in sea surface temperatures or in sea ice cover. Further, the Bern3D+C is a model of intermediate complexity with coarse resolution and does not resolve eddies and other processes operating on small scales. Marine biological production is represented as simple prognostic formulations considering the competition between opal and calcite producers and changes in the uptake ratio of silica to phosphate.

In spite of these limitations, the results of our study allow for a number of robust conclusions. We will argue in the following that the processes represented in our model have probably played a dominant role for reorganizing the global carbon cycle during the early part of the last deglaciation.

### 3.1 Southern Ocean ventilation hypothesis

The “Southern Ocean ventilation hypothesis” (e.g. Francois et al., 1997) is the starting point for the discussion of our modelling results. According to this hypothesis low glacial CO<sub>2</sub> values are related to a poorly ventilated deep ocean and a stably stratified Southern Ocean. Such a configuration would lead to an accumulation of carbon in the deep ocean during glacial periods relative to interglacials. The rise in atmospheric CO<sub>2</sub> over the glacial transition is driven by the break-up of Southern Ocean stratification and the resumption of modern-scale upwelling bringing water rich in carbon, silica, and other nutrients and depleted in <sup>13</sup>C and <sup>14</sup>C to the surface. The characteristics of these

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upwelled water masses are further communicated within the intermediate and surface ocean by entrainment into Subantarctic Mode Water and Antarctic Intermediate Water and to the atmosphere by air-sea gas-exchange. During recent years, a large number of observational evidence has been published supporting the view that changes in SO deepwater formation and in deep ocean ventilation have played an important role for glacial-interglacial CO<sub>2</sub> variations (e.g. Fischer et al., 2009; Spero and Lea, 2002; Adkins et al., 2002; Anderson et al., 2009; Muscheler et al., 2004; Brovkin et al., 2007; Hodell et al., 2003; Marchitto et al., 2007; Francois et al., 1997).

In the following, we will discuss paleo-data from sediment analyses and from ice cores which indicate that large changes in Southern Ocean ventilation indeed occurred during the early part of the deglaciation, roughly in the interval between 18 ka BP and the onset of the Antarctic Cold reversal and the Bølling/Allerød Northern Hemisphere warming at about 14.6 ka BP (Severinghaus and Brook, 1999). We will then compare the data with our model results to constrain the magnitude of the change in atmospheric CO<sub>2</sub> over the last termination which is driven by a SO ventilation mechanism.

### Changes in Southern Ocean opal sedimentation

Anderson et al. (2009) report reconstructed opal fluxes in Southern Ocean sediments which show a rapid increase starting roughly together with the rise in atmospheric CO<sub>2</sub> around 17 ka BP, followed by a deglacial maximum and a long-term decrease into the Holocene (Fig. 8). Opal flux rates are typically several-fold higher during the transition compared to glacial and late-Holocene values. Anderson et al. (2009) interpret these opal flux changes as variations in Southern Ocean upwelling concluding that there was a rapid increase in the upwelling of silica and carbon rich deep water to the Southern Ocean surface at the onset of deglaciation (~18 ka BP). They propose a direct link between the increased ventilation of deep water to the deglacial rise in atmospheric CO<sub>2</sub>.

The step-wise strengthening of SO wind stress as applied in our ocean ventilation experiments drives an immediate rise in deep upwelling in the Southern Ocean in

response to more vigorous deep convection in the model. These simulations may thus serve as model scenarios for an abrupt resumption of upwelling in the SO. Figure 8 shows that our model results for strengthened SO wind stress are in good qualitative agreement with the sediment data of Anderson et al. (2009) in relative magnitude and timing: Increasing SO wind stress by 80% drives a rapid 30%-rise in opal export as more silica is supplied to the surface. Additionally, opal sedimentation also increases and reaches a peak value with a delay of several millennia relative to the reinvigoration of deep upwelling in the Southern Ocean. The peak sedimentation flux is two to three times higher than the initial flux. Both opal export and sedimentation slowly decrease towards initial rates with a typical time scale of 20 000 years. Note however that simulated opal burial fluxes are about 10 times smaller than the reconstructed opal flux (Fig. 8). This is likely due to the spatial averaging over the model grid cell and the relatively low simulated global burial rate of opal as compared to available estimates (Table 1).

Anderson et al. (2009) suggest, based on the positive correlation between  $^{231}\text{Pa}/^{230}\text{Th}$  and opal flux in their records, that the deglacial maximum in opal flux reflects past changes in opal production and Si supply rather than variable opal preservation. In our model however, a relatively modest increase in global silica export of 30% enhances the net flux to the sediments in a highly non-linear way as pore water concentrations of silicic acid and thus the preservation of opal in the sediments greatly increase under the enhanced rain of silica particles (Archer et al., 2000; Ridgwell et al., 2002). The majority of the simulated change in opal sedimentation flux thus reflects increased preservation rather than increased opal production as suggested by Anderson et al. (2009).

Why does the reconstructed opal sedimentation flux relax towards glacial conditions over the course of the Holocene, whereas atmospheric  $\text{CO}_2$  and, presumably, Southern Ocean upwelling remains at high interglacial values? From a continued large Southern Ocean upwelling, one might expect a continued silica supply to the surface ocean, and a sustained high export and sedimentation rate. In our simulation, increased opal

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burial drives a loss of whole-ocean silica. As a result, the simulated export of opal and its sedimentation rate relax towards initial conditions to balance riverine input with the burial flux of silica.

Is the evolution of the opal sedimentation flux directly correlated with opal export and the resumption of upwelling or are there any system time lags? The simulated peak in opal sedimentation is delayed by several millennia compared to the onset of the ventilation in our model. This suggests that the temporal evolution of the ventilation history may be difficult to disentangle exactly from the temporal evolution of the opal sedimentation flux. Nevertheless, the opal records appear consistent with a change in deep ocean ventilation that occurred during the early deglacial period and that may or may not have been abrupt.

### Changes in the stable carbon isotope ratio $\delta^{13}\text{C}$

Figure 9 displays the currently available  $\delta^{13}\text{C}_{\text{CO}_2}$  records from Antarctic ice cores covering the period of the last deglaciation (Smith et al., 1999; Lourantou et al., 2010; Elsig, 2009) together with reconstructed atmospheric  $\text{CO}_2$  (Monnin et al., 2001) and  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  (Stenni et al., 2001) which is a proxy for local temperature on the ice sheet. The  $\delta^{13}\text{C}_{\text{CO}_2}$  records consistently show an early deglacial decrease of 0.3–0.4‰ to reach a minimum at around 15 kaBP and a millennial-scale long-term increase by the same magnitude to attain approximately constant values at 7 ka BP (Elsig et al., 2009). However, the records disagree on the details of the  $\delta^{13}\text{C}_{\text{CO}_2}$  evolution during the Bølling/Allerød-Younger Drias climate swings.

Sediment cores from different ocean basins (Indo-Pacific, sub-Antarctic, and South Atlantic) reveal that the occurrence of carbon isotope minima at the beginning of glacial terminations is a widespread feature in planktic foraminifera  $\delta^{13}\text{C}$  records (Spero and Lea, 2002). The decrease in the  $\delta^{13}\text{C}$  records occurred simultaneously with the initiation of Southern Ocean warming. The data suggests a decrease in surface water  $\delta^{13}\text{C}$  in the early deglacial period (~20 ka BP to 16 ka BP), an absolute minimum at the same

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time as the ice core record of Smith et al. (1999) which is followed by an increase over the rest of the deglacial period into the early Holocene. Taken together, both sediment and ice core data suggest decreasing atmospheric  $\delta^{13}\text{C}$  during the first stage of the deglacial period followed by a long-term increase of similar magnitude.

How can this millennial-scale  $\delta^{13}\text{C}$  evolution be explained? Our results lend support for the interpretation of Spero and Lea (2002) that the early deglacial  $\delta^{13}\text{C}$  decrease reflected renewed Circumpolar Deep Water upwelling with an enhanced transport of low  $\delta^{13}\text{C}$ . A decrease in atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  of 0.3‰ would require a rejuvenation of deep ocean waters by 300 to 400 years according to the  $\delta^{13}\text{C}_{\text{CO}_2}$  sensitivity found in our ventilation experiments (Fig. 6b). Subsequently, this negative signal is reduced by roughly 50% over the following 15 000 yrs due to excess sedimentation of organic material (Fig. 7g). This would leave an increase of 0.1–0.2‰ to be explained by other processes such as carbon uptake by the terrestrial biosphere, surface ocean warming or sea ice retreat.

### Changes in $\text{CaCO}_3$ saturation horizon depth

Analyses of the sedimentary bulk composition indicate a period of enhanced preservation of  $\text{CaCO}_3$  deep sea sediments between ~15–7 ka BP (Ridgwell et al., 2003; Broecker et al., 2001). Broecker et al. (2001) attribute this sedimentary signal to an early deglacial restoration of the terrestrial biosphere. The resulting removal of carbon from the atmospheric and oceanic reservoirs is made responsible for a deepening of the  $\text{CaCO}_3$  saturation horizon between about 15–10 ka BP. The deepening is followed by a gradual shoaling until the latter part of the Holocene which is attributed to carbonate compensation. Ridgwell et al. (2003) invoke regrowth of coral reefs as additional factor driving the deglacial signal of the Pacific  $\text{CaCO}_3$  saturation horizon to simultaneously account for the evolution of atmospheric  $\text{CO}_2$ .

In response to stimulated upwelling in the Southern Ocean rejuvenating deep waters by 300–400 years our model predicts a similar temporal evolution for the Pacific  $\text{CaCO}_3$

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saturation horizon. We simulate a deepening saturation horizon by 300–400 m during the first ~2 kyr and a slow shoaling thereafter to reach initial depths after around 28 kyr (Fig. 7b). The simulated initial deepening is a result of the flushing of the old carbon-rich deep water mass from the Pacific basin and of excess organic matter sedimentation (Fig. 4b). However, the timing of our model results does not match the observations very precisely, maybe due to the unrealistic abrupt change in model boundary conditions. Nevertheless, the agreement is sufficient in order to indicate that increasing deep ocean ventilation and early deglacial excess sedimentation of organic matter may have significantly contributed to the recorded carbonate ion signal in the Pacific ocean in addition to the changes in terrestrial biosphere and growth of coral reefs as proposed by Broecker et al. (2001) and Ridgwell et al. (2003).

### Changes in radiocarbon signatures

Is an increase in deep ocean ventilation consistent with marine radiocarbon data? To generate the full amplitude of the early deglacial 35 ppm increase in atmospheric CO<sub>2</sub>, a decrease in the deep ocean ventilation age of about 300–400 years is required according to the model-derived linear relationship between CO<sub>2</sub> and ventilation age (Fig. 6a). Corresponding simulated changes in deep ocean  $\Delta^{14}\text{C}_{\text{DIC}}$  are an increase of 40–45‰ in the deep North Atlantic of 10–15‰ in the deep North Pacific and in the deep Southern Ocean (Table 3). Atmospheric  $\Delta^{14}\text{C}$  would drop by 30–40‰ (Fig. 6c). The simulated atmosphere-deep ocean  $\Delta^{14}\text{C}$ -difference would thus be reduced by 70–80‰ in the deep Atlantic and by 40–50‰ in the deep North Pacific and the deep Southern Ocean.

A decrease in deep ocean ventilation age of 300–400 years during the early deglacial period seems broadly consistent with available proxy records. However, the locally large changes between 18 ka BP and 14.5 ka BP in the benthic-planktonic age-difference of ~1000 years in the Atlantic sector of the SO (Skinner et al., 2010) or of ~700 years in the deep North Pacific (Galbraith et al., 2007) are not reproduced in our model.

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Our interpretation is further supported by a number of prominent paleo-proxy signals that are adequately reproduced in our simulations. As a response to an abrupt rejuvenation of deep ocean watermasses by 300–400 years which is stimulated by more vigorous deep convection in the Southern Ocean we simulate a decline in  $\delta^{13}\text{C}_{\text{CO}_2}$  by  $\sim 0.3\text{‰}$  during the first 2 kyr followed by a gradual rise as observed in the paleo-records. Taken at face value, the ice core data suggest a slope of 12 ppm per 0.1‰ change in  $\delta^{13}\text{C}_{\text{CO}_2}$  (35 ppm per 0.3‰) during the early deglacial period, whereas the model-derived slope amounts to 13 ppm per 0.1‰. The model response also agrees well with increased opal sedimentation rates in the Southern Ocean during the early deglaciation (Anderson et al., 2009) and is further consistent with the reconstructed evolution of the Pacific  $\text{CaCO}_3$  saturation horizon (Broecker et al., 2001). We finally predict a drop of 30–40‰ in atmospheric  $\Delta^{14}\text{C}_{\text{CO}_2}$  as another result of the invigorated deep ocean ventilation.

Although the simulated model response fits rather well with these specific paleo-proxy records, a perfect match should not be expected. Modeled carbon cycle sensitivities might be subject to considerable uncertainty, in particular due to unsettled questions related to the diffusivity of the oceans (Archer et al., 2000a). It further turned out difficult to reproduce the exact temporal evolution of  $\delta^{13}\text{C}$  and  $\Delta^{14}\text{C}$  as recorded in sediment cores in the model. These signals might sensitively depend upon regional and temporal details of the deglacial reorganization of global ocean circulation and may be further affected by the variations of AMOC (McManus et al., 2004) not considered here. Nonetheless, based on the multi-proxy comparison between our simulated response and the available paleo-data we conclude that accelerating deep ocean ventilation linked to physical processes in the SO has been the dominant driver of the early deglacial changes in global carbon cycling. Early deglacial ocean ventilation changes have continued to significantly contribute to variations in atmospheric  $\text{CO}_2$  until the latter part of the Holocene since it takes millennia for marine sedimentation to re-establish equilibrium with weathering input.

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The glacial-interglacial CO<sub>2</sub> variations appear to be driven by a complex spatio-temporal pattern of processes involving relaxation time-scales of years to millennia. Our analysis of the time-dependent response in the coupled ocean-sediment system highlights that the benefit of conventional "time slice" approaches to study these processes might be rather limited. Our model results suggest that proxy signals arising from deglacial ventilation changes are sustained on a multi-millennial time-scale until the balance between carbon, nutrient and isotope input through rivers and winds and the loss through sedimentation is re-established. Thus, differences in these proxies between the LGM and the late Holocene time slice may partly reflect imbalances in the weathering-sedimentation cycles. This may also be relevant for proxies not included here such as silica or nitrogen isotopes.

We have identified a feedback between the marine nutrient inventory and sedimentation of organic matter. The mechanism emerges from the fact that marine biological productivity in the long run is limited by the riverine input of nutrients into the ocean. Our results indicate that this nutrient-sediment feedback might be important for glacial-interglacial changes in global carbon cycling. For instance, invigorated ocean circulation promotes export production and excess sediment burial of organic matter. The implications for the combined ocean-atmosphere carbon cycle are similar to those of a postulated early deglacial restoration of the terrestrial biosphere (Broecker et al., 2001).

In this study, we have analyzed the time-dependent response of atmospheric CO<sub>2</sub> and its isotopic signatures only to the specific mechanisms of deep ocean ventilation changes and biological pump reorganizations. To improve the process-related understanding of the CO<sub>2</sub>, δ<sup>13</sup>C<sub>CO<sub>2</sub></sub> and Δ<sup>14</sup>C<sub>CO<sub>2</sub></sub> records over the last deglaciation further mechanisms such as sea surface temperature changes, coral reef growth, changes in the carbon stock on land and the spatio-temporal details of the various regional oceanic reorganizations should be considered in coupled three-dimensional global carbon cycle models. Simultaneous model-data comparison on the basis of multiple wisely chosen proxy records seems to be a promising approach towards a better understanding

of the dynamical interplay between the multiple factors driving glacial-interglacial CO<sub>2</sub> fluctuations. A particular challenge is to identify and simulate the physical mechanisms responsible for the changes in Southern Ocean ventilation.

## Appendix A

### Sediment diagenesis model

The sediment diagenesis component used in this study is built on previous work by Heinze et al. (1999) and Gehlen et al. (2006) where the basic modelling concept is based on Archer et al. (1993). The spatial model domain is restricted to the diagenetical zone of the sediments which is assumed here to be the top 10 cm of the surface sediments. Any solid material leaving this domain disappears into the subjacent diagenetically consolidated zone. The interfaces of the 10 model layers are defined at depths of 0, 0.3, 0.6, 1.1, 1.6, 2.1, 3.1, 4.1, 5.1, 7.55, and 10.0 cm. The fraction of pore water volume (porosity) decreases from 0.95 to 0.75, following observations from Ullman and Aller (1982). Model geometry remains fixed during integration.

The vertical movement of solid material in the sediment column is based on the equilibrium: burial = rain – redissolution. When the particulate rain onto the sediment is larger than the redissolution of solid material within the diagenetical zone, the entire column is shifted downwards and sediment burial occurs. Conversely, whenever redissolution is larger than rain, gaps of solid material are produced within the model geometry. These are filled with clay from below. Lateral movements in the sediments are not considered.

Modeled tracers are the four solid components (CaCO<sub>3</sub>, opal, POM and clay) and the six pore water substances (DIC, total alkalinity, phosphate, nitrate, oxygen and silicic acid). CaCO<sub>3</sub> is only considered in the form of calcite. The less stable aragonite form is assumed to be only of secondary importance for the global sediments (Berner and Honjo, 1981). All the solid components are taken to have equal densities ( $\rho = 2.6 \text{ g/cm}^3$ ).

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Considered processes occurring during sediment diagenesis are the dissolution of opal and  $\text{CaCO}_3$  as well as the oxydation and denitrification of POM. The mixing of sediments by burrowing animals such as clams and worms (bioturbation) is parametrized as a diffusion of solid components with closed boundaries at the sediment-water interface and at the bottom of the bioturbated zone.

The mathematical formulation of the sediment model consists of two sets of dynamical equations, one for the four solid components and one for the six pore water substances, along with adequate boundary conditions. The two sets of equations are coupled to each other through the four reaction rates of the sediment-pore water processes (redissolution/remineralization). All of these reactions follow first order kinetics with spatially uniform reaction rate constants. The model parameters of the sediment component are listed in Table A1 along with their standard values and references.

### A1 Redissolution/remineralization

The rate of opal dissolution  $R_{\text{opal}} \left[ \frac{\text{mol Si}}{\text{m}^3 \cdot \text{yr}} \right]$  is proportional to the amount of opal in the sediment,  $c_{\text{opal}}$ , the pore water undersaturation with respect to silicic acid,  $([\text{Si}_{\text{sat}}] - [\text{Si}])$ , and the rate constant,  $r_{\text{opal}}$ :

$$R_{\text{opal}} = r_{\text{opal}} \cdot c_{\text{opal}} \cdot \Phi \cdot (1 - \Phi) \cdot \frac{\rho}{M_{\text{opal}}} \cdot ([\text{Si}_{\text{sat}}] - [\text{Si}]) \cdot H([\text{Si}_{\text{sat}}] - [\text{Si}]). \quad (\text{A1})$$

$M_{\text{opal}}$  stands for the molar weight of opal,  $\rho$  for the density,  $\Phi$  for porosity and  $c_{\text{opal}}$  corresponds to the weight-fraction of opal sediments.  $H(x)$  is the Heaviside step function, preventing spontaneous precipitation of solid opal in case of pore water supersaturation. The silicic acid saturation concentration,  $[\text{Si}]_{\text{sat}}$ , is a spatially constant model parameter set to  $800 \frac{\mu\text{mol}}{\text{l}}$ .

Since  $[\text{Ca}^{2+}]$  is fairly constant and has a relatively high background concentration,  $\text{CaCO}_3$  preservation is primarily governed by the concentration of carbonate ions in the pore water. The rate of  $\text{CaCO}_3$  dissolution  $R_{\text{CaCO}_3} \left[ \frac{\text{mol C}}{\text{m}^3 \cdot \text{yr}} \right]$  is thus linearly related to

the pore water undersaturation with respect to carbonate ions,  $([\text{CO}_3]_{\text{sat}} - [\text{CO}_3])$ , the weight-fraction of  $\text{CaCO}_3$ ,  $c_{\text{CaCO}_3}$ , and the rate constant,  $r_{\text{CaCO}_3}$ :

$$R_{\text{CaCO}_3} = r_{\text{CaCO}_3} \cdot c_{\text{CaCO}_3} \cdot \Phi \cdot (1 - \Phi) \cdot \frac{\rho}{M_{\text{CaCO}_3}} \quad (\text{A2})$$

$$\times ([\text{CO}_3]_{\text{sat}} - [\text{CO}_3]) \cdot H([\text{CO}_3]_{\text{sat}} - [\text{CO}_3]),$$

$$[\text{CO}_3]_{\text{sat}} = \frac{[\text{Ca}^{2+}]}{K'_{\text{sp}}}. \quad (\text{A3})$$

The solubility product of calcite  $K'_{\text{sp}}$  is calculated after Mucci (1983) with a pressure dependence following Millero (1995). The calcium ion concentration  $[\text{Ca}^{2+}]$  is proportional to salinity.

The rate of organic matter oxydation  $R_{\text{oxy}} \left[ \frac{\text{mol C}}{\text{m}^3 \cdot \text{yr}} \right]$  is limited by the availability of oxygen in the pore water,  $[\text{O}_2]$ , and by the amount of POC,  $c_{\text{POC}}$ , in the sediment:

$$R_{\text{oxy}} = r_{\text{oxy}} \cdot c_{\text{POC}} \cdot \Phi \cdot (1 - \Phi) \cdot \frac{\rho}{M_{\text{POC}}} \cdot \frac{\mathcal{R}_C}{\mathcal{R}_{\text{O}_2}} \cdot [\text{O}_2] \cdot H([\text{O}_2]). \quad (\text{A4})$$

The rate constant for oxydation,  $r_{\text{oxy}}$ , is set to  $50 \frac{\text{l}}{\text{mol}} \cdot \text{yr}$ .  $\mathcal{R}_C$  and  $\mathcal{R}_{\text{O}_2}$  are the Redfield ratios for carbon and oxygen, respectively.

Organic matter denitrification only occurs in suboxic conditions, i.e. below a critical oxygen concentration in the pore water  $[\text{O}_2]_{\text{crit}}$ . The reaction rate  $R_{\text{denit}} \left[ \frac{\text{mol C}}{\text{m}^3 \cdot \text{yr}} \right]$  is proportional to the POC weight-fraction,  $c_{\text{POC}}$ , the availability of nitrate,  $[\text{NO}_3^-]$ , and the rate constant  $r_{\text{denit}}$ :

$$R_{\text{denit}} = r_{\text{denit}} \cdot c_{\text{POC}} \cdot \Phi \cdot (1 - \Phi) \cdot \frac{\rho}{M_{\text{POC}}} \cdot \frac{\mathcal{R}_C}{\mathcal{R}_{\text{denit}}} \cdot [\text{NO}_3^-] \cdot H([\text{NO}_3^-]) \cdot H([\text{O}_2]_{\text{crit}} - [\text{O}_2]). \quad (\text{A5})$$

$\mathcal{R}_{\text{denit}}$  is the Redfield coefficient for denitrification. The reaction rate,  $r_{\text{denit}}$ , is set to  $50 \frac{\text{l}}{\text{mol} \cdot \text{yr}}$ .

Inert clay is not subject to redissolution in the sediments:

$$R_{\text{clay}} = 0. \quad (\text{A6})$$

## A2 Solid components

The four solid sediment components are subject to bioturbation, advection and remineralization/redissolution. They satisfy the following equations:

$$\frac{\partial c_{\text{CaCO}_3}}{\partial t} = D_B \frac{\partial^2}{\partial z^2} (c_{\text{CaCO}_3}) - \frac{\partial}{\partial z} (w c_{\text{CaCO}_3}) - R_{\text{CaCO}_3} \cdot \frac{M_{\text{CaCO}_3}}{\rho} \cdot \frac{1}{1 - \Phi}, \quad (\text{A7})$$

$$\frac{\partial c_{\text{opal}}}{\partial t} = D_B \frac{\partial^2}{\partial z^2} (c_{\text{opal}}) - \frac{\partial}{\partial z} (w c_{\text{opal}}) - R_{\text{opal}} \cdot \frac{M_{\text{opal}}}{\rho} \cdot \frac{1}{1 - \Phi}, \quad (\text{A8})$$

$$\frac{\partial c_{\text{POC}}}{\partial t} = D_B \frac{\partial^2}{\partial z^2} (c_{\text{POC}}) - \frac{\partial}{\partial z} (w c_{\text{POC}}) - (R_{\text{oxy}} + R_{\text{denit}}) \cdot \frac{M_{\text{POC}}}{\rho} \cdot \frac{1}{1 - \Phi}, \quad (\text{A9})$$

$$\frac{\partial c_{\text{clay}}}{\partial t} = D_B \frac{\partial^2}{\partial z^2} (c_{\text{clay}}) - \frac{\partial}{\partial z} (w c_{\text{clay}}). \quad (\text{A10})$$

$c_S$  corresponds to the dry weight fraction of  $\text{CaCO}_3$ , opal, POC or clay.  $D_B$  is the bioturbation coefficient,  $w$  is the vertical advection velocity and  $R_x$  are the reaction rates. The variable porosities are taken into account when solving Eqs.(A7)–(A10).

Below the bioturbated zone, the sediment composition remains unchanged:

$$\left. \frac{\partial c_S}{\partial z} \right|_{z \geq 10 \text{cm}} = 0, \quad \text{for each solid component } S. \quad (\text{A11})$$

At the sediment-water interface ( $z = 0 \text{ cm}$ ) the vertical advection velocity is given by the sum of the volume rain rates,  $\text{Rain}_S \cdot M_S / \rho$  [ $\frac{\text{cm}^3}{\text{cm}^2 \cdot \text{yr}}$ ], divided by the solid sediment fraction,  $(1 - \Phi(0))$ :

$$w(0) = \frac{1}{1 - \Phi(0)} \sum_{\text{solid components}} \text{Rain}_S \cdot M_S / \rho. \quad (\text{A12})$$

Vertical advection velocities,  $w(z)$ , decrease with depth due to remineralization and redissolution as well as due to increasing solid sediment fractions:

$$w(z) < w(0), \text{ for } z > 0. \quad (\text{A13})$$

The velocity  $w(z)$  is given by the remaining volume flux at depth  $z$  divided by the solid sediment fraction,  $(1 - \Phi(z))$ :

$$w(z) = \frac{1}{1 - \Phi(z)} \left( \sum_{\text{solid components}} \text{Rain}_S \cdot M_S / \rho - \int_{z'=0}^z \sum_{\text{solid components}} R_S(z') \cdot M_S(z') / \rho \cdot dz' \right). \quad (\text{A14})$$

The burial flux of the solid component  $S$  at the base of the bioturbated zone ( $z = 10 \text{ cm}$ ),  $F_S^{\text{burial}}$  [ $\frac{\text{mol}}{\text{cm}^2 \cdot \text{yr}}$ ], is given by:

$$F_S^{\text{burial}} = w(10) \cdot c_S(10) \cdot \frac{\rho}{M_S} \cdot (1 - \Phi(10)). \quad (\text{A15})$$

In case total redissolution is larger than total rain in the sediment column, gaps of solid material are produced within the fixed model geometry. These are filled with clay in order to ensure mass conservation:

$$C_{\text{opal}} + C_{\text{cal}} + C_{\text{org}} + C_{\text{clay}} = 1. \quad (\text{A16})$$

### A3 Pore water solutes

The concentrations of pore water solutes are subject to sediment remineralization and redissolution as well as to pore water diffusion. Here, the diffusion in a porous medium is represented by multiplying the uniform pore water diffusion coefficient  $D$  by the porosity  $\Phi$ . Experimentally derived diffusion coefficients do not deviate much from these values (Ullman and Aller, 1982). The dynamical equations for the pore water solutes are:

$$\frac{\partial Si}{\partial t} = \frac{\partial}{\partial z} \left( \Phi D \frac{\partial Si}{\partial z} \right) + R_{\text{opal}} / \Phi, \quad (\text{A17})$$

$$\frac{\partial O_2}{\partial t} = \frac{\partial}{\partial z} \left( \Phi D \frac{\partial O_2}{\partial z} \right) - \frac{R_{O_2}}{R_C} \cdot R_{\text{oxy}} / \Phi, \quad (\text{A18})$$

$$\frac{\partial NO_3^-}{\partial t} = \frac{\partial}{\partial z} \left( \Phi D \frac{\partial NO_3^-}{\partial z} \right) + \frac{R_{NO_3^-}}{R_C} \cdot R_{\text{oxy}} / \Phi - \frac{R_{\text{denit}}}{R_C} \cdot R_{\text{denit}} / \Phi, \quad (\text{A19})$$

$$\frac{\partial PO_4}{\partial t} = \frac{\partial}{\partial z} \left( \Phi D \frac{\partial PO_4}{\partial z} \right) + \frac{R_{PO_4}}{R_C} \cdot (R_{\text{oxy}} + R_{\text{denit}}) / \Phi, \quad (\text{A20})$$

$$\frac{\partial DIC}{\partial t} = \frac{\partial}{\partial z} \left( \Phi D \frac{\partial DIC}{\partial z} \right) + (R_{\text{calc}} + R_{\text{oxy}} + R_{\text{denit}}) / \Phi, \quad (\text{A21})$$

$$\begin{aligned} \frac{\partial ALK}{\partial t} = \frac{\partial}{\partial z} \left( \Phi D \frac{\partial ALK}{\partial z} \right) - \frac{R_{NO_3^-} + 2 \cdot R_{PO_4}}{R_C} \cdot R_{\text{oxy}} / \Phi \\ + \frac{R_{\text{denit}} - 2 \cdot R_{PO_4}}{R_C} \cdot R_{\text{denit}} / \Phi + 2 \cdot R_{\text{cal}} / \Phi. \end{aligned} \quad (\text{A22})$$

The boundary condition for pore water solutes at the ocean-sediment interface is:

$$C|_{z=0 \text{ cm}} = C_{\text{bottomwater}}, \text{ for each solute tracer } C. \quad (\text{A23})$$

Below the bioturbated zone the solute concentrations are assumed to remain constant:

$$\left. \frac{\partial C}{\partial z} \right|_{z \geq 10 \text{ cm}} = 0, \text{ for each solute tracer } C. \quad (\text{A24})$$

The pH-equilibrium in the pore water is given by:

$$\frac{[CO_2][CO_3^{=}]}{[HCO_3^-]^2} = \frac{K_1'}{K_2'}. \quad (\text{A25})$$

The apparent dissociation constants  $K_1'$  and  $K_2'$  are calculated after Mehrbach et al. (1973) with pressure dependence following Millero (1995).

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**Table 1.** Bulk numbers characterising the preindustrial steady state of the coupled ocean-sediment Bern3D+C model compared to previous modelling results and independent estimates.

Variable	Units	this study	Heinze et al. (2003)	Ind. estimates
atm. CO <sub>2</sub>	ppm	278.0	277.7	–
POC export production	GtC yr <sup>-1</sup>	12.76	8.66	6.5–13.1 <sup>a</sup>
POC rain onto sediments	GtC yr <sup>-1</sup>	0.49	0.12	1.7–3.3 <sup>a</sup>
POC accumulation	GtC yr <sup>-1</sup>	0.14	0.048	0.12–0.26 <sup>a</sup>
CaCO <sub>3</sub> export production	GtC yr <sup>-1</sup>	1.13	1.64	0.8–1.2 <sup>b</sup>
CaCO <sub>3</sub> rain onto sediments	GtC yr <sup>-1</sup>	0.40	0.42	0.5 <sup>c</sup>
CaCO <sub>3</sub> accumulation	GtC yr <sup>-1</sup>	0.17	0.18	0.1–0.14 <sup>d</sup>
Opal export production	Tmol Si yr <sup>-1</sup>	95	185	102–178 <sup>b</sup>
Opal rain onto sediments	Tmol Si yr <sup>-1</sup>	64.9	81.3	13–47 <sup>e</sup>
Opal accumulation	Tmol Si yr <sup>-1</sup>	1.12	5.5	5.3–8.9 <sup>e</sup>
POC sediment pool	GtC	428	241.9	–
CaCO <sub>3</sub> sediment pool	GtC	881	1010	–
Opal sediment pool	Tmol Si	9779	19986	–

<sup>a</sup> Sarmiento and Gruber (2006)

<sup>b</sup> Jin et al. (2006)

<sup>c</sup> Millimann and Droxler (1996)

<sup>d</sup> Feely et al. (2004)

<sup>e</sup> Treguer et al. (1995)

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**Table 2.** Ocean ventilation experiments: Simulated changes in ocean biogeochemistry of the coupled ocean-sediment Bern3D+C model at the time slice 2 kyr after the abrupt variation in ocean ventilation.

SO wind stress amplitude	North Atlantic below 2000 m (0°–71° N)				North Pacific below 2000 m (0°–63° N)			
	$\Delta^{14}\text{C}$	$\delta^{13}\text{C}$	$\text{PO}_4$	$\text{O}_2$	$\Delta^{14}\text{C}$	$\delta^{13}\text{C}$	$\text{PO}_4$	$\text{O}_2$
	[‰]	[‰]	[ $\frac{\text{mol}}{\text{m}^3}$ ]	[ $\frac{\text{mol}}{\text{m}^3}$ ]	[‰]	[‰]	[ $\frac{\text{mol}}{\text{m}^3}$ ]	[ $\frac{\text{mol}}{\text{m}^3}$ ]
20%	-260	0.171	2.47	115.4	-273	0.092	2.55	103.9
40%	-258	0.179	2.49	131.5	-265	0.011	2.54	126.0
60%	-195	0.632	1.94	186.4	-279	0.011	2.67	114.7
80%	-188	0.671	1.89	190.1	-279	0.026	2.66	118.5
Control	-180	0.711	1.84	194.2	-278	0.049	2.64	124.0
120%	-173	0.734	1.80	198.4	-275	0.077	2.62	133.5
140%	-168	0.745	1.77	201.0	-274	0.102	2.59	141.1
160%	-163	0.754	1.74	203.5	-273	0.123	2.58	147.8
180%	-156	0.774	1.70	206.8	-272	0.141	2.56	153.1
SO wind stress amplitude	Southern Ocean below 2000 m (South of 51° S)				Southern Ocean surface (South of 51° S)			
	$\Delta^{14}\text{C}$	$\delta^{13}\text{C}$	$\text{PO}_4$	$\text{O}_2$	$\Delta^{14}\text{C}$	$\delta^{13}\text{C}$	$\text{PO}_4$	$\text{O}_2$
	[‰]	[‰]	[ $\frac{\text{mol}}{\text{m}^3}$ ]	[ $\frac{\text{mol}}{\text{m}^3}$ ]	[‰]	[‰]	[ $\frac{\text{mol}}{\text{m}^3}$ ]	[ $\frac{\text{mol}}{\text{m}^3}$ ]
20%	-196	0.385	2.22	156.9	-61	2.350	0.60	290.2
40%	-198	0.377	2.25	174.6	-73	2.198	0.68	289.8
60%	-208	0.308	2.34	171.1	-99	2.049	1.01	289.6
80%	-208	0.330	2.33	175.8	-106	1.944	1.09	289.4
Control	-206	0.358	2.31	182.1	-112	1.871	1.15	288.7
120%	-205	0.392	2.28	192.5	-119	1.799	1.21	288.3
140%	-204	0.425	2.26	201.2	-125	1.725	1.28	287.9
160%	-202	0.455	2.24	209.4	-130	1.658	1.35	287.6
180%	-199	0.490	2.22	217.0	-135	1.603	1.38	287.4

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**Table 3.** Long-term response in atmospheric CO<sub>2</sub> and amplification by sedimentary feedbacks.

SO wind stress amplitude	Global Deep Water Age [years]	Atmospheric CO <sub>2</sub> without sediments after 50 kyr [ppm]	Atmospheric CO <sub>2</sub> with sediments after 50 kyr [ppm]	Sedimentary amplification of the CO <sub>2</sub> response [%]
20%	2097	251	234	63%
40%	2062	252	236	62%
60%	1963	269	265	44%
80%	1908	273	271	40%
Control	1840	278	278	–
120%	1770	285	287	29%
140%	1706	290	294	33%
160%	1640	295	302	30%
180%	1578	299	308	30%

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**Table A1.** Parameters for the sediment component of the Bern3D+C model along with standard values and references.

Symbol	Parameter	Value	Reference
	Number of time steps per year	48	this study
$\rho$	Density of solid sediment components	2.6 g cm <sup>-3</sup>	Heinze et al. (1999)
$M_{\text{POM}}$	Molar weight of POM	32.74 g mol <sup>-1</sup>	Heinze et al. (1999)
$M_{\text{CaCO}_3}$	Molar weight of CaCO <sub>3</sub>	100.0 g mol <sup>-1</sup>	Heinze et al. (1999)
$M_{\text{opal}}$	Molar weight of opal	67.2 g mol <sup>-1</sup>	Heinze et al. (1999)
$M_{\text{clay}}$	Molar weight of clay	430.51 g mol <sup>-1</sup>	Heinze et al. (1999)
	Terrestrial clay flux	1 g m <sup>-2</sup> yr <sup>-1</sup>	Heinze et al. (1999)
$r_{\text{CaCO}_3}$	Rate constant for CaCO <sub>3</sub> redissolution	1000 l mol <sup>-1</sup> yr <sup>-1</sup>	Heinze et al. (1999)
$r_{\text{opal}}$	Rate constant for opal redissolution	20 l mol <sup>-1</sup> yr <sup>-1</sup>	this study
$r_{\text{oxy}}$	Rate constant for oxydation	50 l mol <sup>-1</sup> yr <sup>-1</sup>	Heinze et al. (1999)
$r_{\text{denit}}$	Rate constant for denitrification	50 l mol <sup>-1</sup> yr <sup>-1</sup>	Gehlen et al. (2006)
$[\text{Si}]_{\text{sat}}$	Saturation concentration for pore water silicic acid	800 μmol l <sup>-1</sup>	Heinze et al. (2003)
$D$	Pore water diffusion coefficient	8. × 10 <sup>-6</sup> cm <sup>2</sup> s <sup>-1</sup>	Heinze et al. (1999)
$D_B$	Bioturbation diffusion coefficient	15. × 10 <sup>-3</sup> cm <sup>2</sup> yr <sup>-1</sup>	Heinze et al. (1999)
$[\text{O}_2]_{\text{crit}}$	Critical oxygen concentration for denitrification	1 μmol l <sup>-1</sup>	Gehlen et al. (2006)
$\mathcal{R}_{\text{PO}_4}$	Redfield coefficient for phosphate	1	Anderson and Sarmiento (1994)
$\mathcal{R}_{\text{O}_2}$	Redfield coefficient for oxygen	170	Anderson and Sarmiento (1994)
$\mathcal{R}_{\text{NO}_3^-}$	Redfield coefficient for nitrate	16	Anderson and Sarmiento (1994)
$\mathcal{R}_C$	Redfield coefficient for carbon	117	Anderson and Sarmiento (1994)
$\mathcal{R}_{\text{denit}}$	Redfield coefficient for denitrification	97.6	Gehlen et al. (2006)

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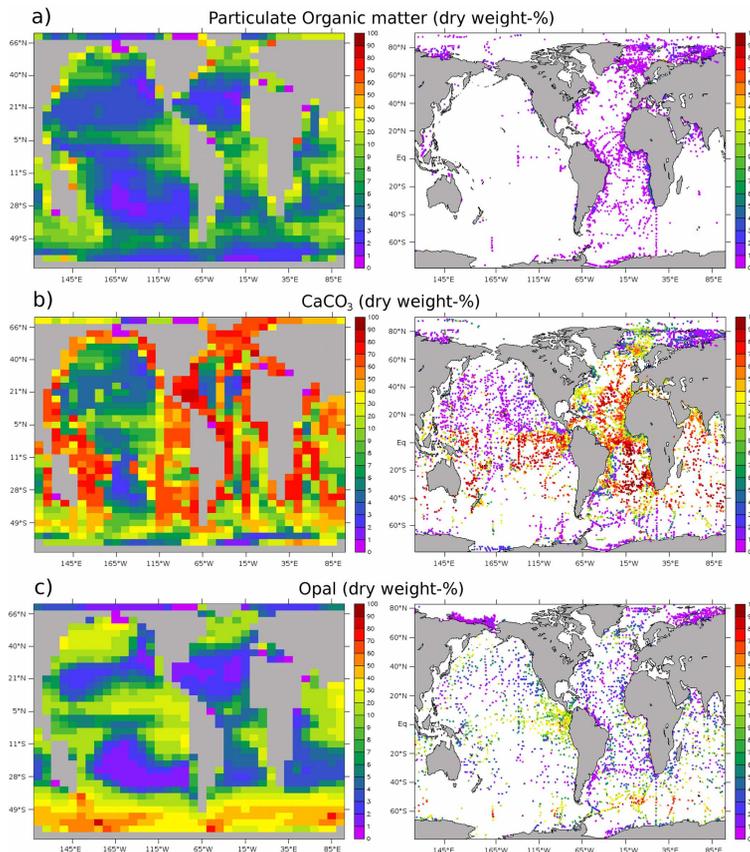
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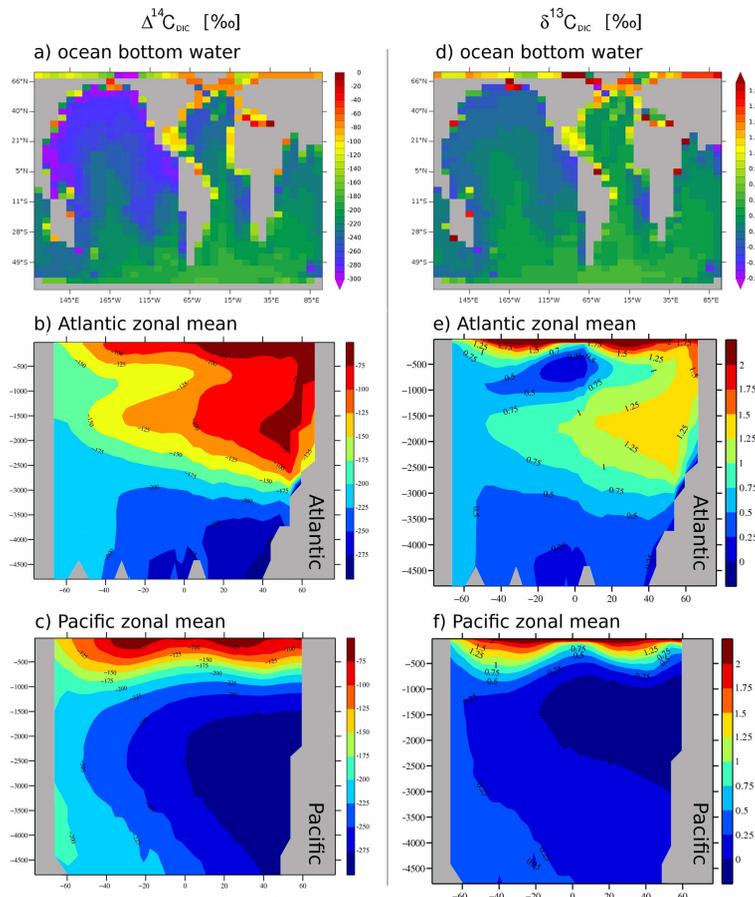
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**Fig. 1.** Modelled (left row) and observed (right row) composition of surface sediments on the ocean bottom: **(a)** Particulate organic matter, **(b)**  $\text{CaCO}_3$  and **(c)** opal. Modelled sediment composition is averaged over the top 1.1 cm of the sediment column. Sediment data is from the Pangaea database (Publishing Network for Geoscientific & Environmental Data, <http://orfois.pangaea.de>). Note that the color scale is not linear.

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**Fig. 2.** Modelled distribution of the isotope signatures in ocean dissolved inorganic carbon for preindustrial boundary conditions.  $\Delta^{14}\text{C}_{\text{DIC}}$  is shown in the left row and  $\delta^{13}\text{C}_{\text{DIC}}$  in the right row. Top panels (a and d) show bottom water signatures, the middle row panels (b and e) display Atlantic and bottom panels (c and f) Pacific zonal mean sections in preindustrial steady state.

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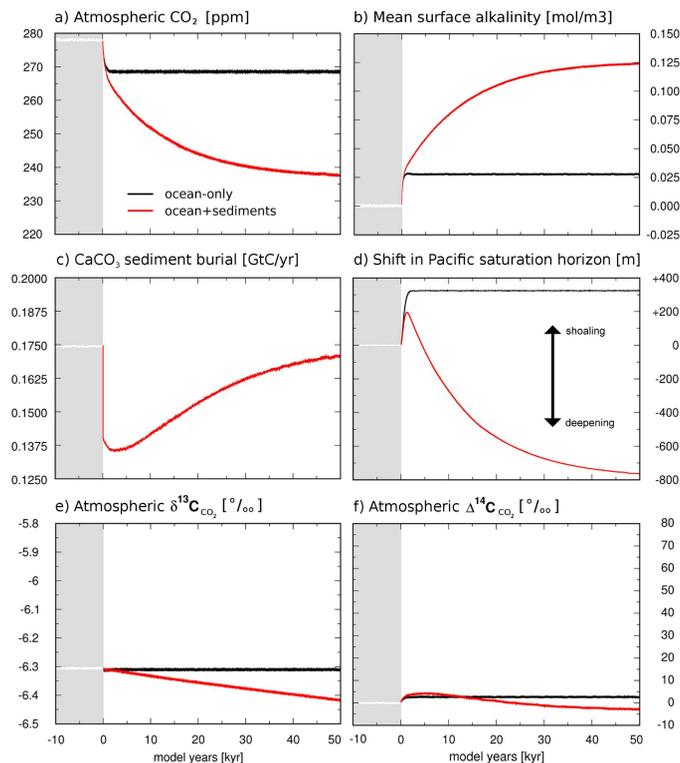
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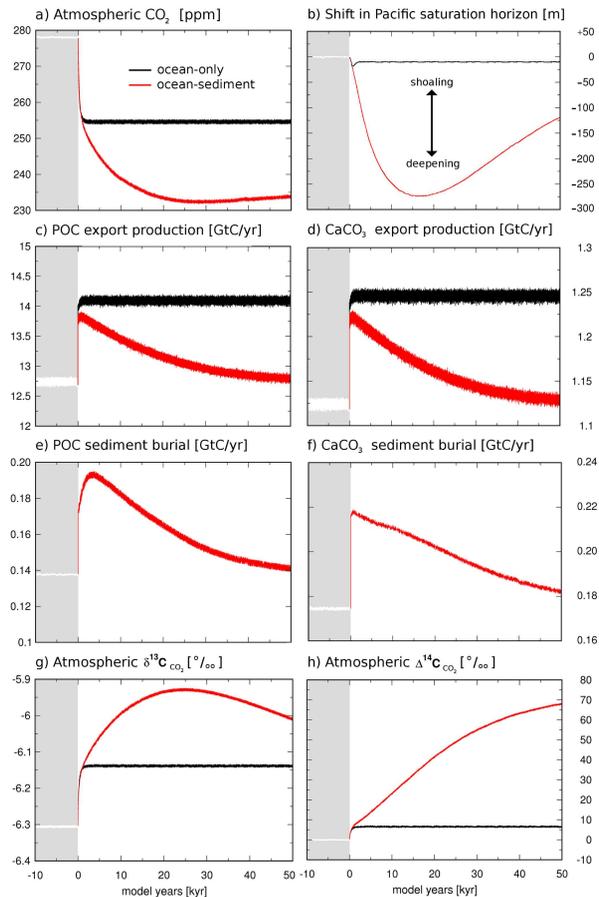
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**Fig. 3.** Rain ratio reduction: Simulated response to a 20%-reduction in  $\text{CaCO}_3$  export. **(a)** atmospheric  $\text{CO}_2$ , **(b)** mean surface alkalinity, **(c)**  $\text{CaCO}_3$  sedimentation, **(d)** shift in Pacific  $\text{CaCO}_3$  saturation horizon, **(e)** atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  and **(f)** atmospheric  $\Delta^{14}\text{C}_{\text{CO}_2}$ .

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**Fig. 4.** Strengthening of the marine biological cycle: Simulated response to an abrupt 30%-increase in the ocean's phosphate content. **(a)** atmospheric  $\text{CO}_2$ , **(b)** shift in Pacific  $\text{CaCO}_3$  saturation horizon, **(c)** POC export production, **(d)**  $\text{CaCO}_3$  export production, **(e)** POC sedimentation, **(f)**  $\text{CaCO}_3$  sedimentation, **(g)** atmospheric  $\delta^{13}\text{C}_{\text{CO}_2}$  and **(h)** atmospheric  $\Delta^{14}\text{C}_{\text{CO}_2}$ .

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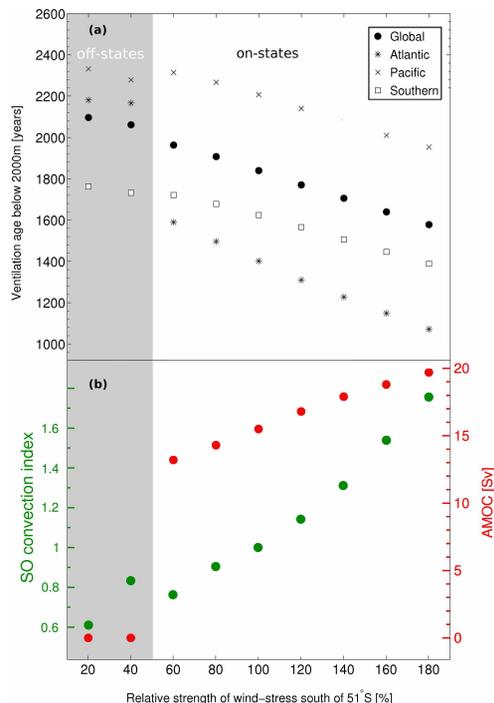
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**Fig. 5. (a)** Ventilation age in different ocean basins as a function of the SO wind stress amplitude. Ventilation age is calculated from simulated  $^{14}R$ -signatures in the ocean below 2000 m depth.  $^{14}R$  is treated here as an ideal age tracer in the sense that it is transported only by ocean circulation and the atmospheric signature is held constant ( $^{14}R_{\text{atm}}=1.0$ ). Ventilation age is calculated as  $\tau_{\text{basin}} = \frac{1}{\lambda} \ln \left( \frac{^{14}R_{\text{atm}}}{\overline{^{14}R_{\text{basin}}}} \right)$  where  $\lambda = 1/8267 \text{ yr}^{-1}$  is the decay constant for  $^{14}C$  and  $\overline{^{14}R_{\text{basin}}}$  is the average  $^{14}R$ -signature in the respective basin.

**(b)** Southern Ocean convection index and Atlantic Meridional Overturning Circulation (AMOC) versus SO wind stress. The SO convection index is proportional to the number of model grid cells in the SO participating in convective mixing.

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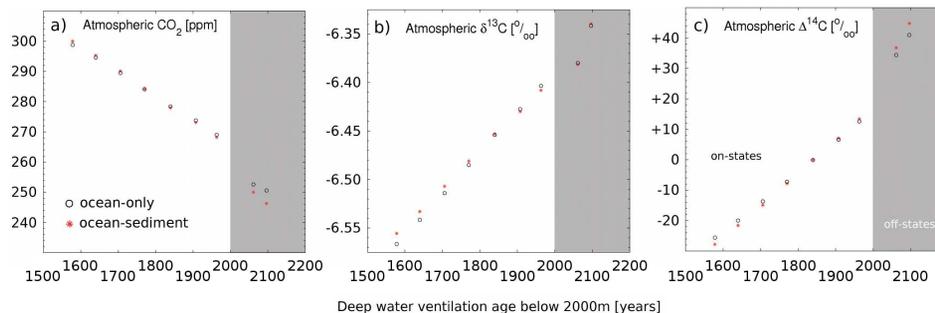
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**Fig. 6.** Response in atmospheric **(a)**  $\text{CO}_2$ , **(b)**  $\delta^{13}\text{C}_{\text{CO}_2}$ , and **(c)**  $\Delta^{14}\text{C}_{\text{CO}_2}$  to changes in ocean ventilation. Displayed values are diagnosed for the time slice at 2 kyr after the perturbation in SO wind forcing. At this time carbon cycling has reached a new equilibrium in the ocean-only model (circles), whereas the ocean-sediment model (stars) is still under relaxation at 2 kyr due to the millennial-scale response of ocean-sediment interactions.

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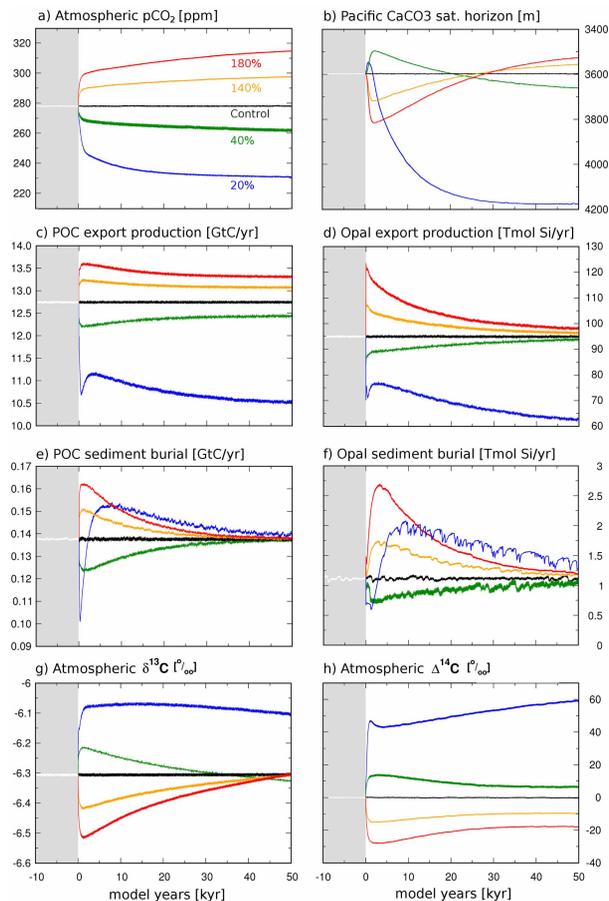
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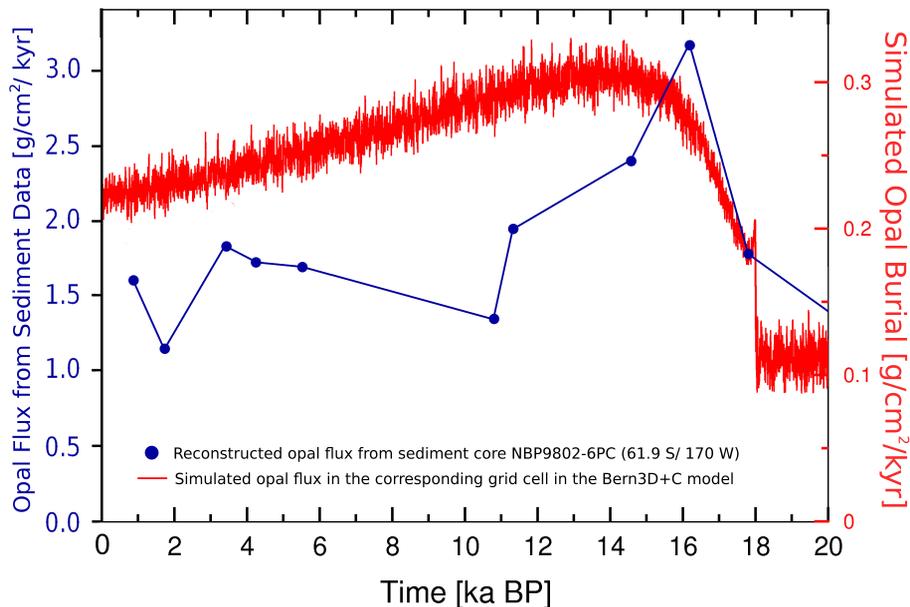
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**Fig. 7.** Ocean ventilation experiments: time-dependent carbon cycle response of the coupled ocean-sediment system to changes in deep ocean ventilation induced by scaling the amplitude of windstress and deep convection in the Southern Ocean.

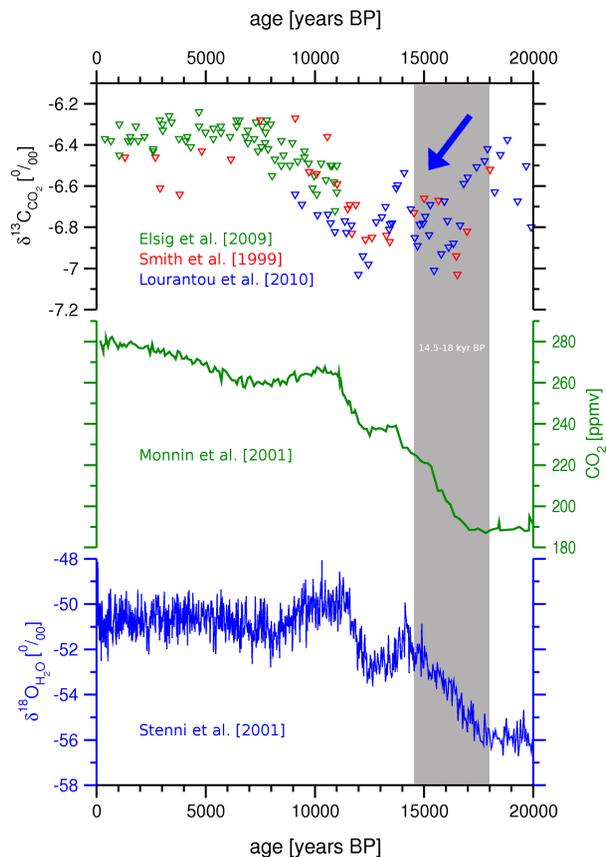
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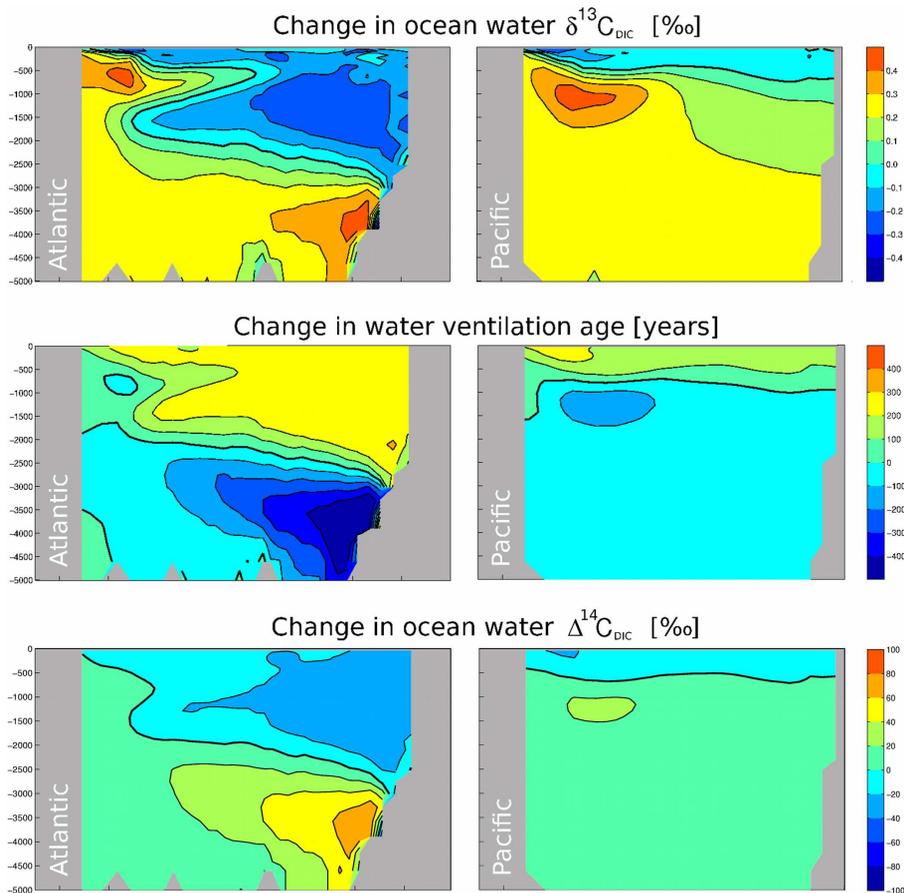


**Fig. 8.** Reconstructed local opal flux (blue line) at 61.9° S/170° W from Anderson et al. (2009) and simulated opal burial rate in the corresponding model cell (red line) as simulated in the experiment with an abrupt 80%-increase in SO wind stress at 18 ka BP.

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**Fig. 9.** Paleo-records from Antarctic ice cores:  $\delta^{13}\text{C}$ -data is from Elsig et al. (2009) (green triangles), Smith et al. (1999) (red triangles) and Lourantou et al. (2010) (blue triangles). The blue arrow indicates the recorded drop in  $\delta^{13}\text{C}_{\text{CO}_2}$  ( $\sim -0.4\%$ ) starting around 18 ka BP.  $\text{CO}_2$ -data is from Monnin et al. (2001) and  $\delta^{18}\text{O}$ -data from Stenni et al. (2001).



**Fig. 10.** Model response for  $\delta^{13}\text{C}_{\text{DIC}}$ , ventilation age and  $\Delta^{14}\text{C}_{\text{DIC}}$  to stimulated deep mixing in the Southern Ocean and to more vigorous and deeper AMOC as a result of a 80%-increase in SO wind stress amplitude. Shown are the simulated differences in Atlantic and Pacific zonal mean sections at 15 ka after the the step-wise perturbation of the model state.